

Part F. Nomenclature of oxyacids – the -ic acids and the -ous acids

Table 7, Classification of types of compounds

Formula of acid	Anion	Name of acid
H_2SO_4	SO_4^{2-} , sulfate	sulfuric acid
HClO_4	ClO_4^- , perchlorate	perchloric acid
H_2SO_3	SO_3^{2-} , sulfite	sulfurous acid
HNO_3	NO_3^- , nitrate	nitric acid
HBrO	BrO^- , hypobromite	hypobromous acid
H_2CO_3	CO_3^{2-} , carbonate	carbonic acid
HNO_2	NO_2^- , nitrite	nitrous acid

Naming Oxyacids (using table 7 above)

1. What's the identity of the cation for each of the acids?
2. What's the trend in the name of the acids of the -ate oxyanions?
3. What's the trend in the name of the acids of the -ite oxyanions?
4. Write a general statement that describes how to predict the formula and name of an oxyacid given the formula and name of the oxyanion.

Naming Oxyacids Answer Sheet

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