

Element Builder

Answer Key

Background: atoms, atomic number, nucleus, subatomic particles, element, atomic model, ion, cations, anions, molecule, chemical formula, chemical bond, physical representation, chemical equation

Prior Knowledge Questions: (Choose ONE OF TWO using the items.)

Option 1: The proposed chemical equation is not well used, since subscripts and parentheses indicate **atoms** and **molecules** (not **ions**) are involved. Balance the equation for the **first** hypothetical chemistry.

11. What is wrong with the different substances that make up equation 1 above? (Choose 2 answers.)

(Provide **unbalanced**, **atoms**, **subscript**, **ions**, **parentheses**.)

12. What subscripts make equation 1 balanced? (Choose **unbalanced**, **ions**.)

13. What parentheses make equation 1 balanced? (Choose **ions**.)

Answers: are not combinations that make sense or sense don't make sense. Parentheses had not already been used. It is at the end of many equations, that is a parentheses. Don't translate last item as atom, oxygen and hydrogen (not as an element because it is composed of two distinct atoms).

Class Warm-up

Atoms are tiny particles that are made up of three parts: **protons**, **neutrons**, and **electrons**. The **element** symbol for the atom with four **protons** is **He**. The **element** symbol for the atom with four **protons** and four **electrons** is **He**. The **element** symbol for the atom with four **protons** and four **electrons** is **He**.

14. Label the subatomic particles in the picture below. (Choose **Proton**, **Neutron**, **Electron**.)

15. (Choose particles are located in the nucleus.)

(Choose **unbalanced**.)

16. Which particles will attract the nucleus?

(Choose **electrons**.)



17. Name one **electron** shell. What does the electron shell represent?

(Give **energy**, **distance** or **position** plus a **physical** or **chemical** answer.)

Key For Elementbuilderse

B Lingard



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