#### AP Chemistry Practice Test Chapter 8 and 9

#### Multiple Choice

Identify the choice that best completes the statement or answers the question.

13	Which pair of atoms should form the most polar bond	9
	a. F and B d. N	
		and N
	c. F and O	
23	Which pair of ions should form the ionic lattice with the highest energy?	
	a. Na <sup>1+</sup> and Br <sup>1-</sup> d. Li	1* and O2-
	b. Li <sup>1*</sup> and F <sup>1-</sup> c. K	1+ and F1-
	c. Cs <sup>1+</sup> and F <sup>1-</sup>	
 3.	The electrostatic force of attraction is greatest in which compound?	
	a. BaO d. M	
	b. MgO e. C.	aO
	c. CaS	
4.	4. Which molecule has the weakest bond?	
	a, CO d. N	
	b. O <sub>2</sub> e. C	l <sub>2</sub>
	c. NO	
5.	5. For which species are octet resonance structures neces	
		eF <sub>2</sub>
	b. SO <sub>2</sub> c. So	D <sub>4</sub> -
	c. CO <sub>2</sub>	
 6.	6. How are the bonding pairs arranged in the best Lewis	
	a. O-O-O d. O	
	b. O=O-O e. O	=O=O
	c. O=O-O	
7.	7. Which species has the shortest bond length?	
	a. CN <sup>1</sup> - d. St	$O_1$ :
	b. O <sub>2</sub> e. O	0,
	e. SO <sub>2</sub>	
8.	8. Which species requires the least amount of energy to	remove an electron from the outermost energy level?
100	a. Na <sup>1*</sup> d. O	
		g <sup>2</sup> *
	c. F1-	
9.	Which species has a valid non-octet Lewis structure?	
		eCl <sub>4</sub>
	b. SiF <sub>4</sub> e. O	O <sub>2</sub> -

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