

Name _____
SPC _____

The Atom and Nuclear Chemistry Review

1. Neutrons are neutral, but protons are _____ and electrons are _____.
2. a) Francium-212: How many neutrons? _____ How many protons? _____
b) $^{63}_{27}\text{Cu}$: How many neutrons? _____ How many protons? _____
c) Give the symbol and the chemical name for an atom with 20 protons and 22 neutrons.
3. What was the first element found to be radioactive? _____
4. What is happening when an element is radioactive? _____
5. What role do the neutrons play in keeping the nucleus of an atom stable? _____
6. Elements with more than _____ protons in their nucleus are always radioactive.
7. Define background radiation. _____
8. What are some sources of naturally occurring radiation? _____
9. What are some human-made sources of radiation? _____
10. What is an alpha particle? _____

Write its symbol: _____
11. What is a beta particle? _____

Write its symbol: _____
12. Which is more harmful, an alpha particle or a beta particle? Why? How can each be stopped? _____
13. Where do alpha and beta particles come from? _____
14. When a beta particle forms, a _____ atom turns into a _____ atom.

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AN Whitehead



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