


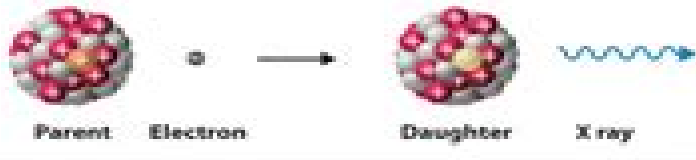

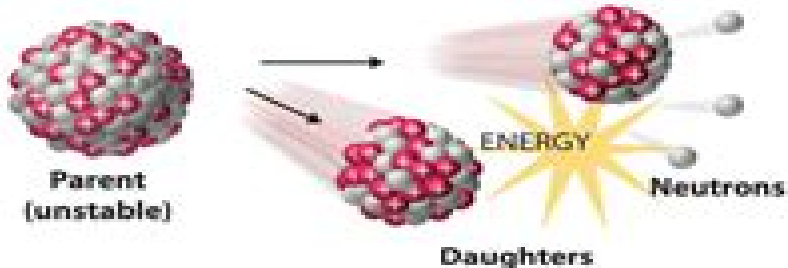


Decay Type	Radiation Emitted	Generic Equation	Model		
Alpha decay	$\frac{4}{2}\alpha$	$\frac{A}{Z}X \longrightarrow \frac{A-4}{Z-2}X' + \frac{4}{2}\alpha$		Parent	Daughter Alpha Particle
Beta decay	$\frac{0}{-1}\beta$	$\frac{A}{Z}X \longrightarrow \frac{A}{Z+1}X' + \frac{0}{-1}\beta$		Parent	Daughter Beta Particle
Positron emission	$\frac{0}{+1}\beta$	$\frac{A}{Z}X \longrightarrow \frac{A}{Z-1}X' + \frac{0}{+1}\beta$		Parent	Daughter Positron
Electron capture	X rays	$\frac{A}{Z}X + \frac{0}{-1}e \longrightarrow \frac{A}{Z-1}X' + \text{X ray}$		Parent Electron	Daughter X ray
Gamma emission	$\frac{0}{0}\gamma$	$\frac{A}{Z}X^* \xrightarrow{\text{Relaxation}} \frac{A}{Z}X' + \frac{0}{0}\gamma$		Parent (excited nuclear state)	Daughter Gamma ray
Spontaneous fission	Neutrons	$\frac{A+B+C}{Z+Y}X \longrightarrow \frac{A}{Z}X' + \frac{B}{Y}X' + C\frac{1}{0}n$		Parent (unstable)	Daughters ENERGY Neutrons

Key Terms Radioactivity And Nuclear Reactions

Ralph H. Petrucci



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sense of chemistry as a field that not only has a lively history but also one that is currently dynamic with important new developments on the horizon Fundamentals of Chemistry Elizabeth P. Rogers,1987 Foundations of College Chemistry Morris Hein,Susan Arena,Cary Willard,2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity University Physics Hugh D. Young,Roger A. Freedman,2000 This work aims to enable students to develop physical intuition and strong problem solving skills In addition it points out the conceptual and computational pitfalls that commonly plague beginner physics students **Introduction to Chemical Principles** Howard Stephen Stoker,1986 **Physical Science** Bill W. Tillery,1999 An introductory text for a one or two semester text on the fundamental behavior of matter and energy for nonscience majors covering physics chemistry astronomy and earth sciences Material incorporates minimal use of technical terminology discussion of applications and environmental concerns throughout and historical background with emphasis on problem solving mathematics Learning aids include chapter summaries key terms and multiple choice and critical thinking questions Exercises are given in two sets one with explained answers Includes an art notebook of explanatory color diagrams Annotation copyright by Book News Inc Portland OR

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