

Name: \_\_\_\_\_

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## Chemistry 11

## Classifying Matter

## Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements  
Mixtures  
Compound  
Symbols  
Properties

Carbon  
Pure  
Letters  
Carbon dioxide

- a. Elements cannot be separated by physical or chemical means.
- b. Elements are abbreviated with symbols, which consist of one or two letters.
- c. Two or more elements chemically combined make up a compound.
- d. Carbon is an example of an element and carbon dioxide is an example of a compound.
- e. Elements and compounds are called pure substances because they have a unique set of chemical and physical properties.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	<u>Na</u>	<u>+1</u>
b. Copper	<u>Cu</u>	<u>+2, +1</u>
c. Tungsten	<u>W</u>	<u>+6</u>
d. Iron	<u>Fe</u>	<u>+3, +2</u>
e. Tin	<u>Sn</u>	<u>+4, +2</u>
f. Potassium	<u>K</u>	<u>+1</u>
g. Gold	<u>Au</u>	<u>+3, +1</u>
h. Silver	<u>Ag</u>	<u>+1</u>

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a.  $\text{P}^{3-}$  I

b.  $\text{H}_2\text{O}$  M

c.  $\text{O}_2$  M

d.  $\text{Au}^{3+}$  I

e. Ge A

f. O A

g.  $\text{Ca}^{2+}$  I

h.  $\text{NH}_3$  M

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