

Name: _____ School: _____

THE PERIODIC TABLE

- (a) The Periodic Table of elements was put together in 1869 by The Russian chemist known as Dmitri Mendeleev.
- (b) The elements in The Periodic Table are arranged into three main categories, which are **metals**, **nonmetals** and **semi metals**.
- Names and symbols of first 20 elements.
 - Refer to Periodic table for the other elements.

NAMES AND SYMBOLS							
1. Hydrogen	H	6. Carbon	C	11. Sodium	Na	16. Sulfur	S
2. Helium	He	7. Nitrogen	N	12. Magnesium	Mg	17. Chlorine	Cl
3. Lithium	Li	8. Oxygen	O	13. Aluminum	Al	18. Argon	Ar
4. Beryllium	Be	9. Fluorine	F	14. Silicon	Si	19. Potassium	K
5. Boron	B	10. Neon	Ne	15. Phosphorus	P	20. Calcium	Ca

- (c) Periodic means the properties of the elements have a repeating nature.
- (d) The properties of elements repeat chemically and physically, e.g. **Sodium**, **Lithium** and **Potassium** are white in colour, soft, change colour in seconds if exposed to air.
- (e) The symbol for an element is either a **capital letter** or a **capital letter followed by a lower case letter**.
- (f) Elements in the periodic table are arranged in the order of their increasing **atomic numbers**.
- (g) Elements in the same period have the same number of electron shells or energy levels.
- (h) Dmitri Mendeleev used atomic masses instead of atomic numbers to arrange elements.

Name	nitrogen	Atomic number
	7	
	N	Symbol
Atomic mass	14.007	

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