#### MATTER AND MATERIAL

#### **GRADE 9**

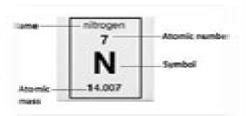
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#### THE PERIODIC TABLE

- (a) The Periodic Table of elements was put together in 1865 by The Russian chemist known as Dmitri Mendeleev.
- (b) The elements in The Periodic Table are arranged into three main categories, which are metals, nonmetals and semi metals.
- Names and symbols of first 20 elements.
- Refer to Periodic table for the other elements.

NAMES AND SYMBOLS									
1. Hydrogen	н	6. Carbon	C	11.	Sodium	Na	16. Sulfur	- 5	
2. Hellum	Hie	7. Nitrogen	N	12.	Magnesium	Mg	17. Chlorine	CI	
3. Lithium	LI	8. Oxygen	0	13.	Aluminum	,A,I	18. Argon	Arr	
4. Beryllium	Вe	9. Fluorine	F	1-4,	Silicon	57	15. Potassium	180	
5. Boron	В	10.Neon	Nie	1.5.	Phosphorus	p	20. Calcium	Ca	

- (c) Periodic means the properties of the elements have a repeating nature.
- (d) The properties of elements repeat chemically and physically, e.g. Sodium, Lithium and Potasium are white in colour, soft, change colour in seconds if exposed to air.
- (e) The symbol for an element is either a capital letter or a capital letter followed by a lower case letter.
- (f) Elements in the periodic table are arranged in the order of their increasing atomic numbers.
- (g) Elements in the same period have the samenumber of electron shells or energy levels.
- (h) Dmitri Mendeleev used atomic masses instead of atomic numbers to arrange elements.



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