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## Student Exploration: Limiting Reactants

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**Vocabulary:** chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript.

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a Wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make? \_\_\_\_\_
2. Which ingredient limited the number of hot dogs you could make? \_\_\_\_\_
3. Which ingredient will you have leftovers of? \_\_\_\_\_


### Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the Limiting Reactants Gizmo™, you can determine which reactant is limiting in various scenarios.



To begin, make sure  $\text{H}_2 + \text{O}_2$  becomes  $\text{H}_2\text{O}$  is selected. The small "2" in  $\text{H}_2$ ,  $\text{O}_2$ , and  $\text{H}_2\text{O}$  is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of  $\text{O}_2$  molecules and  $\text{H}_2$  molecules to two.
  - A. How many hydrogen molecules ( $\text{H}_2$ ) are there? \_\_\_\_\_ Hydrogen atoms? \_\_\_\_\_
  - B. How many oxygen molecules ( $\text{O}_2$ ) are there? \_\_\_\_\_ Oxygen atoms? \_\_\_\_\_
2. How many  $\text{H}_2\text{O}$  molecules do you think will form when these four molecules react? \_\_\_\_\_
3. Click **Play** () . How many  $\text{H}_2\text{O}$  molecules actually formed? \_\_\_\_\_

# Limiting Reactants Answer Key

**John A. Olmsted, Gregory M. Williams**

## Limiting Reactants Answer Key:

**Kaplan SAT Subject Test Chemistry 2015-2016** Kaplan Test Prep, 2015-03-03 Essential strategies practice and review to ace the SAT Subject Test Chemistry Getting into a top college has never been more difficult Students need to distinguish themselves from the crowd and scoring well on a SAT Subject Test gives students a competitive edge Kaplan's SAT Subject Test Chemistry is the most up to date guide on the market with complete coverage of both the content review and strategies students need for success on test day Kaplan's SAT Subject Test Chemistry features A full length diagnostic test Full length practice tests Focused chapter summaries highlights and quizzes Detailed answer explanations Proven score raising strategies End of chapter quizzes Kaplan is serious about raising students scores we guarantee students will get a higher score

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**General Chemistry Key Concise Review Guide Sixth Edition** Darrell D. Ebbing, 1999 *Holt Chemistry* Ralph Thomas Myers, 2004 *eBook: General, Organic and Biological Chemistry 2e* SMITH, 2012-02-16 *eBook General Organic and Biological Chemistry 2e*

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**Preparatory Chemistry** Howard Stephen Stoker, 1990

**Student Solutions Manual: Ssm Chemistry** Deborah Wiegand, 2003 This manual contains complete worked out solutions to all follow up problems and about half of all the chapter problems Each chapter of solutions opens with a summary of the text chapter content and a list of key equations needed to solve the problems

**Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook** McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A

Chemistry Handbook Matter and Change      SourceBook Version 2.1 ,1998      **Prentice Hall Chemistry** ,2000

**General Chemistry** Ralph H. Petrucci, William S. Harwood, 1993 General Chemistry Principles and Modern Applications is recognized for its superior problems lucid writing and precision of argument This updated and expanded edition retains the popular and innovative features of previous editions including Feature Problems follow up Integrative and Practice Exercises to accompany every in chapter Example and Focus On application boxes as well as new Keep in Mind marginal notes Topics covered include atoms and the atomic theory chemical compounds and reactions gases Thermochemistry electrons in atoms chemical bonding liquids solids and intermolecular forces chemical kinetics principles of chemical equilibrium acids and bases electrochemistry representative and transitional elements and nuclear and organic chemistry For individuals interested in a broad overview of chemical principles and applications      Must Know High School Chemistry Mary Millhollon, Richard H. Langley, 2019-06-28 A UNIQUE NEW APPROACH THAT S LIKE A LIGHTNING BOLT TO THE BRAIN You know that moment when you feel as though a lightning bolt has hit you because you finally get something That s how this book will make you react We hope Each chapter makes sure that what you really need to know is clear right off the bat and sees to it that you build on this knowledge Where other books ask you to memorize stuff we re going to show you the must know ideas that will guide you toward success in chemistry You will start each chapter learning what the must know ideas behind a chemistry subject are and these concepts will help you solve the chemistry problems that you find in your classwork and on exams Dive into this book and find 250 practice questions that mirror what you will find in your classwork and on exams A bonus app with flashcards that will reinforce what you ve learned Extensive examples that drive home essential concepts An easy access setup that allows you to jump in and out of subjects Chemistry topics aligned to national and state education standards Special help for more challenging chemistry subjects including the mole concept stoichiometry and solutions We re confident that the must know ideas in this book will have you up and solving chemistry problems in no time or at least in a reasonable amount of time      *Holt Chemistry File* ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket      **Study Guide to Accompany Basics for Chemistry** Martha Mackin, 1982-01-01 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides

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