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Malaguti Ciak 50 E1 E2 Service Manual

RJ Alexander



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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1/267 per degree in 1874 Victor Regnault obtained better experimental results showing this number to be 1/273 and concluded that the pressure would approach zero at 1/273 degrees C

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22 414 l we can substitute 101 325 kPa for pressure 22 414 l for volume and 273 15 K for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure (STP) for reporting properties of gases 273 15 K and 1 atm 101 325 kPa 1 atm STP one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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