

Answers to the Student book

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1 States of matter

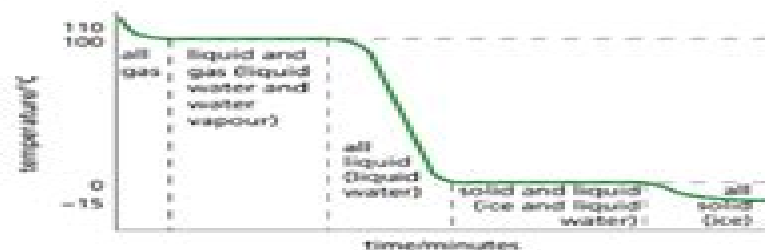
Test Yourself Questions

Page 4

- 1 The particles in copper are vibrating about a fixed point. As the temperature increases, they gain kinetic energy and vibrate more, colliding with neighbouring particles. This has the effect of increasing the overall volume of the metal. In other words, it expands.

Page 7

- 2 The particles in gases have much weaker forces of attraction between their particles as the particles are already further away from each other compared with those in a solid. These weaker forces of attraction are more easily broken as temperatures rise, allowing the particles to move away from one another more easily.
- 3 There is friction between the ice and the air as the car moves through it. This friction produces heat which causes the ice to melt.
- 4 Adding salt to ice lowers the freezing point of the ice. As long as the temperature is not lower than the freezing point of the mixture of salt and ice, the ice will melt.
- 5



The graph should show that when the changes of state occur the temperature does not change until the change of state is complete. So there should be a flat line at 100°C as the steam condenses to form water and at 0°C as the water freezes to form ice.

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Rachel S Tattersall



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Table of Contents Nature Approach To Chemistry Answer Key Page 46

1. Understanding the eBook Nature Approach To Chemistry Answer Key Page 46
 - The Rise of Digital Reading Nature Approach To Chemistry Answer Key Page 46
 - Advantages of eBooks Over Traditional Books
2. Identifying Nature Approach To Chemistry Answer Key Page 46
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nature Approach To Chemistry Answer Key Page 46
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nature Approach To Chemistry Answer Key Page 46
 - Personalized Recommendations
 - Nature Approach To Chemistry Answer Key Page 46 User Reviews and Ratings
 - Nature Approach To Chemistry Answer Key Page 46 and Bestseller Lists
5. Accessing Nature Approach To Chemistry Answer Key Page 46 Free and Paid eBooks

- Nature Approach To Chemistry Answer Key Page 46 Public Domain eBooks
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 - Nature Approach To Chemistry Answer Key Page 46 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Nature Approach To Chemistry Answer Key Page 46
 - Highlighting and Note-Taking Nature Approach To Chemistry Answer Key Page 46
 - Interactive Elements Nature Approach To Chemistry Answer Key Page 46
8. Staying Engaged with Nature Approach To Chemistry Answer Key Page 46
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nature Approach To Chemistry Answer Key Page 46
9. Balancing eBooks and Physical Books Nature Approach To Chemistry Answer Key Page 46
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nature Approach To Chemistry Answer Key Page 46
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Nature Approach To Chemistry Answer Key Page 46
- Setting Reading Goals Nature Approach To Chemistry Answer Key Page 46
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Nature Approach To Chemistry Answer Key Page 46
- Fact-Checking eBook Content of Nature Approach To Chemistry Answer Key Page 46
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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