SAMPLE PAPER (2023 -24)

CHEMISTRY THEORY (043)

Max. Marks:70

Time: 3 hours

General Instructions:

Read the following instructions carefully.

- (a) There are 33 questions in this question paper with internal choice.
- (b) SECTION A consists of 16 multiple -choice questions carrying 1 mark each.
- (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
- (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
- (e) SECTION D consists of 2 case based questions carrying 4 marks each.
- (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
- (q) All questions are compulsory.
- (h) Use of log tables and calculators is not allowed.

SECTION A

The following questions are multiple -choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.

- 1. Which of the following solutions will have the highest conductivity at 298 K?
 - (a) 0.01 M HCl solution

(b) 0.1 M HCl solution

(c) 0.01 M CH 3COOH solution

(d) 0.1 M CH 3COOH solution

Identify A and B:

- (a) A = 1-phenylethanal, B = acetophenone
- (b) A = Benzophenone B = formaldehyde
- (c) A= Benzaldehyde, B = Acetophenone
- (d) A = Benzophenone, B = Acetophenone
- The vitamins which can be stored in our body are:
 - (a) Vitamin A, B, D and E

(d) Vitamin A, C, D and K

(c) Vitamin A, B, C and D

- (d) Vitamin A, D, E and K
- 4. What is IUPAC name of the ketone A, which undergoes iodo form reaction to give

CH3 CH= C(CH3)COONa and yellow precipitate of CHI3?

(a) 3-Methylpent-3-en-2one

(b) 3-Methylbut-2-en- one

(c) 2, 3-Dimethylethanone

(d) 3-Methylpent-4-one

- 5. Which of the following is not correct?
 - (a) In haloarenes, the electron pairs on halogen atom are in conjugation with π -electrons of the ring.
 - (b) The carbon-magnesium bond is covalent and non-polar in nature .
 - (c) During S_N1 reaction, the carbocation formed in the slow step being sp2 hybridised is planar.
 - (d) Out of CH2= CH-Cl and C6H3CH2Cl, C6H3CH2Cl is more reactive towards SN1 reaction

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- 1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to June 2chemistry 970paper 3question
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 - Chapter 3: June 2chemistry 970paper 3question in Everyday Life
 - ∘ Chapter 4: June 2chemistry 970paper 3question in Specific Contexts
 - ∘ Chapter 5: Conclusion
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- 4. In chapter 3, this book will examine the practical applications of June 2chemistry 970paper 3question in daily life. The third chapter will showcase real-world examples of how June 2chemistry 970paper 3question can be effectively utilized in everyday scenarios.
- 5. In chapter 4, this book will scrutinize the relevance of June 2chemistry 970paper 3question in specific contexts. This chapter will explore how June 2chemistry 970paper 3question is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about June 2chemistry 970paper 3question. The final chapter will summarize the key points that have been discussed throughout the book.
 - The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of June 2chemistry 970paper 3question.

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