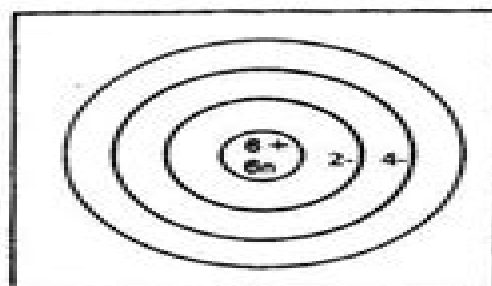
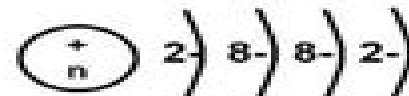


# MAKING ATOM MODELS

Draw models of the atoms of the elements indicated in the squares below. The mass numbers and atomic numbers are given for each element. Represent protons with a + sign, neutrons with an n, and electrons with small circles on the rings that represent the energy levels. The model of a carbon atom has been completed for you as an example. Use only the energy levels that are necessary.

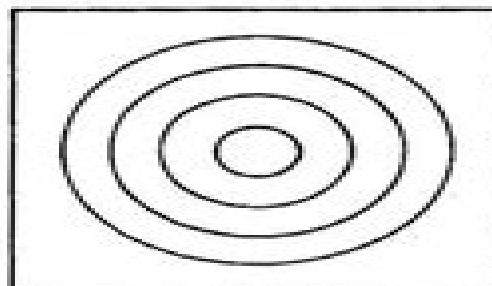
You will need to check the dot diagrams using the picture link



**CARBON**

Atomic number = 6  
Mass number = 12

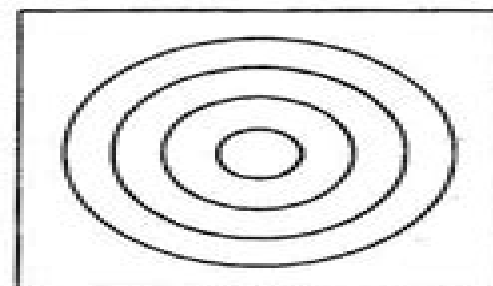
Protons \_\_\_\_\_  
Neutrons \_\_\_\_\_  
Electrons \_\_\_\_\_  
Family \_\_\_\_\_  
Period \_\_\_\_\_



**NEON**

Atomic number = 10  
Mass number = 20

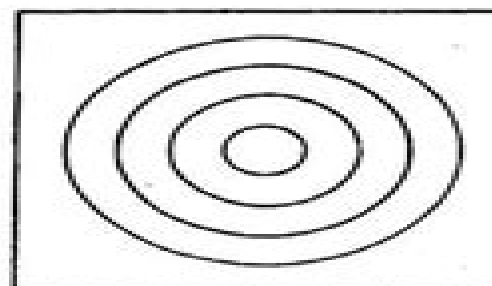
Protons \_\_\_\_\_  
Neutrons \_\_\_\_\_  
Electrons \_\_\_\_\_  
Family \_\_\_\_\_  
Period \_\_\_\_\_



**SODIUM**

Atomic number = 11  
Mass number = 23

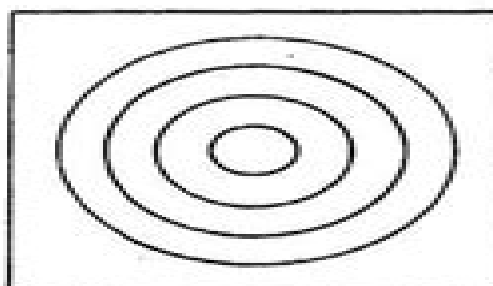
Protons \_\_\_\_\_  
Neutrons \_\_\_\_\_  
Electrons \_\_\_\_\_  
Family \_\_\_\_\_  
Period \_\_\_\_\_



**CHLORINE**

Atomic number = 17  
Mass number = 35

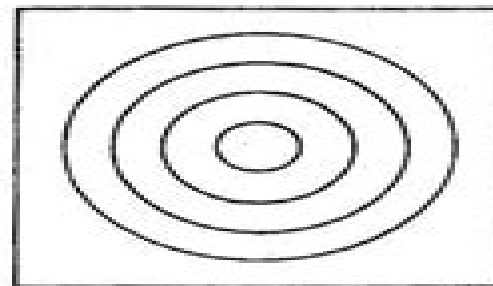
Protons \_\_\_\_\_  
Neutrons \_\_\_\_\_  
Electrons \_\_\_\_\_  
Family \_\_\_\_\_  
Period \_\_\_\_\_



**OXYGEN**

Atomic number = 8  
Mass number = 16

Protons \_\_\_\_\_  
Neutrons \_\_\_\_\_  
Electrons \_\_\_\_\_  
Family \_\_\_\_\_  
Period \_\_\_\_\_



**MAGNESIUM**

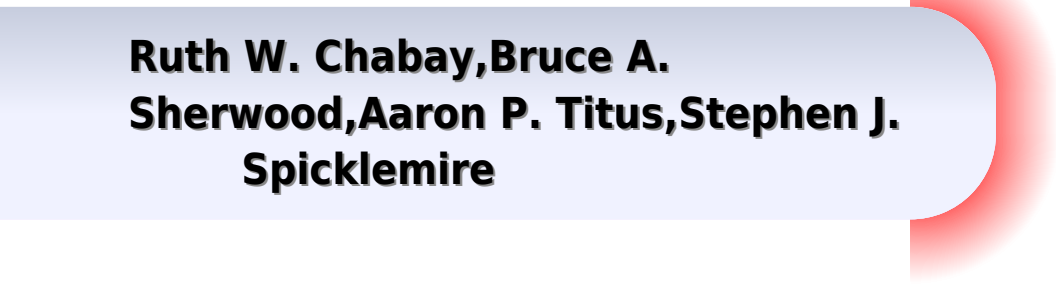
Atomic number = 12  
Mass number = 24

Protons \_\_\_\_\_  
Neutrons \_\_\_\_\_  
Electrons \_\_\_\_\_  
Family \_\_\_\_\_  
Period \_\_\_\_\_



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**Ruth W. Chabay, Bruce A.  
Sherwood, Aaron P. Titus, Stephen J.  
Spicklemire**



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**International Tables for Crystallography, Volume F** Eddy Arnold, Daniel M. Himmel, Michael G.

Rossmann, 2012-03-05 International Tables for Crystallography Volume F is an expert guide to macromolecular crystallography for the structural biologist. It was commissioned by the International Union of Crystallography in recognition of the extraordinary contributions that knowledge of macromolecular structure has made and will make to the analysis of biological systems from enzyme catalysis to the workings of a whole cell. The volume covers all stages of a crystallographic analysis from the preparation of recombinant proteins through crystallization, diffraction data collection, phase determination, structure validation and structure analysis. Although the volume is written for experienced scientists, it is recognized that the reader is more likely to be a biologist interested in structure than a classical crystallographer interested in biology. Thus there are chapters on the fundamentals, history and current perspectives of macromolecular crystallography as well as on useful programs and databases such as the Protein Data Bank. Each chapter is written by one or more internationally recognized experts. This second edition features 19 new articles and many articles from the first edition have been revised. The new articles cover topics such as standard definitions for quality indicators, expression of membrane proteins, protein engineering, high throughput crystallography, radiation damage, merohedral twinning, low resolution *ab initio* phasing, robotic crystal loading, whole cell X-ray diffraction imaging and halogen interactions in biological crystal structures. There are also new articles on relevant software including software for electron microscopy. These enhancements will ensure that Volume F continues to be a key reference for macromolecular crystallographers and structural biologists. More information on the series can be found at <http://it.iucr.org>.

**Matter and Interactions, Volume 2** Ruth W. Chabay, Bruce A. Sherwood, Aaron P. Titus, Stephen J. Spicklemire, 2025-03-18 Matter and Interactions Volume 2: Electric and Magnetic Interactions, 5th Edition. Matter Interactions is a calculus-based introductory physics text that reflects a modernized view of physics. It stresses reasoning from powerful physics principles and integrates contemporary insights such as the atomic nature of matter, quantized energy, and relativistic dynamics throughout the curriculum. Students engage in the full process of creating and refining physical models. Computational modeling is integrated to allow students to apply fundamental principles to more complex realistic systems and to explore the possible ranges of behavior of physical models. Joining Ruth Chabay and Bruce Sherwood for this edition as authors are longtime collaborators Aaron Titus, North Carolina State University, and Stephen Spicklemire, University of Indianapolis, who have made great impacts on the new video series, interactive figures, and simulations. The new edition is thoughtfully updated with extensive content revisions, including chapter and section-level learning objectives clarified and simplified, initial presentation of key concepts and techniques, and the introduction of angular momentum earlier, before collisions.

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Science Tutor: Chemistry, Grades 7 - 8 Raham, 2008-08-28. Connect students in grades 7 and up with science using Science Tutor Chemistry. This effective 48 page resource provides additional concept reinforcement for students who struggle in chemistry. Each lesson in this book contains an Absorb section to instruct and simplify concepts and an Apply section to help students grasp concepts on their own. The book covers topics such as matter, physical and chemical changes, mixtures and solutions, the periodic table, atomic structure, and radioactivity. It is great for use in the classroom and at home.

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