

## Chemical Equations and Reactions

### SECTION 2

**SHORT ANSWER** Answer the following questions in the space provided.

1. Match the equation type on the left to its representation on the right.

\_\_\_ **c** \_\_\_ synthesis



\_\_\_ **d** \_\_\_ decomposition



\_\_\_ **b** \_\_\_ single-displacement



\_\_\_ **a** \_\_\_ double-displacement



2. \_\_\_ **c** \_\_\_ In the reaction described by the equation  $2Al(s) + 3Fe(NO_3)_2(aq) \rightarrow 3Fe(s) + 2Al(NO_3)_3(aq)$ , iron has been replaced by

(a) nitrate.

(c) aluminum.

(b) water.

(d) nitrogen.

3. \_\_\_ **a** \_\_\_ Of the following chemical equations, the only reaction that is both synthesis and combustion is



4. \_\_\_ **b** \_\_\_ Of the following chemical equations, the only reaction that is both combustion and decomposition is



(d)

5. Identify the products when the following substances decompose:

a. **Its separate elements** \_\_\_\_\_ a binary compound

b. **Metal oxide + water** \_\_\_\_\_ most metal hydroxides

c. **Metal oxide + carbon dioxide** \_\_\_\_\_ a metal carbonate

d. **Water + sulfur dioxide** \_\_\_\_\_ the acid  $H_2SO_3$

6. The complete combustion of a hydrocarbon in excess oxygen yields the products

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## **Table of Contents Mix Review Chemical Equations And Reactions Answers**

1. Understanding the eBook Mix Review Chemical Equations And Reactions Answers
  - The Rise of Digital Reading Mix Review Chemical Equations And Reactions Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Mix Review Chemical Equations And Reactions Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mix Review Chemical Equations And Reactions Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mix Review Chemical Equations And Reactions Answers
  - Personalized Recommendations
  - Mix Review Chemical Equations And Reactions Answers User Reviews and Ratings
  - Mix Review Chemical Equations And Reactions Answers and Bestseller Lists
5. Accessing Mix Review Chemical Equations And Reactions Answers Free and Paid eBooks
  - Mix Review Chemical Equations And Reactions Answers Public Domain eBooks
  - Mix Review Chemical Equations And Reactions Answers eBook Subscription Services
  - Mix Review Chemical Equations And Reactions Answers Budget-Friendly Options

6. Navigating Mix Review Chemical Equations And Reactions Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Mix Review Chemical Equations And Reactions Answers Compatibility with Devices
  - Mix Review Chemical Equations And Reactions Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mix Review Chemical Equations And Reactions Answers
  - Highlighting and Note-Taking Mix Review Chemical Equations And Reactions Answers
  - Interactive Elements Mix Review Chemical Equations And Reactions Answers
8. Staying Engaged with Mix Review Chemical Equations And Reactions Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mix Review Chemical Equations And Reactions Answers
9. Balancing eBooks and Physical Books Mix Review Chemical Equations And Reactions Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mix Review Chemical Equations And Reactions Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Mix Review Chemical Equations And Reactions Answers
  - Setting Reading Goals Mix Review Chemical Equations And Reactions Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mix Review Chemical Equations And Reactions Answers
  - Fact-Checking eBook Content of Mix Review Chemical Equations And Reactions Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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