Section 4.5

Types of Aqueous Solutions and Solubility

Solubility Rules for Ionic Compounds in Water

Compounds Containing the Following Ions Are Generally Soluble	Exceptions
Li*, Na*, K*, and NH ₄ *	None
NO ₃ ⁻ and C ₂ H ₃ O ₂ ⁻	None
Off, 8rf, and Iff	When these ions pair with ${\rm Ag}^+, {\rm Hg_2}^{2+},$ or ${\rm Pb}^{2+},$ the resulting compounds are insoluble.
SO ₄ 2-	When $SO_4{}^2$ pairs with $Sr^2{}^+$, $Ba^2{}^+$, $Pb^2{}^+$, $Ag^4{}_+$, or $Ca^2{}^+$, the resulting compound is insoluble.
Compounds Containing the Following Ions Are Generally Insoluble	Exceptions
OH" and S ² "	When these ions pair with Li $^+$, Na $^+$, K $^+$, or NH $_e{}^+$ the resulting compounds are soluble.
	When S ² pairs with Ca ²⁺ , Sr ²⁺ , or Ba ²⁺ , the resulting compound is soluble.
	When OH " pairs with Ca2", Sr2", or Ba2", the resulting compound is slightly soluble.
CO ₃ 2- and PO ₄ 3-	When these ions pair with Li+, Na+, K+, or NH _a + the resulting compounds are soluble.

Model Rules Of Solubility In Aqueous Solutions Pogil

John Henry Saylor

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Handbook of Aqueous Solubility Data Samuel H. Yalkowsky, Yan He, Parijat Jain, 2016-04-19 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until the first publication of the Handbook of Aqueous Solubility Data this information had been scattered throughout numerous sources Now newly revised the second edition of **Handbook of Aqueous Solubility Data** Samuel H. Yalkowsky, 2003-03-26 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until now this information has been scattered throughout the literature Containing over 16 000 solubility data points for more than 4 000 organic compounds Handbook of Aqueous **The Experimental Determination of Solubilities** G. T. Hefter, R. P. T. Tomkins, 2003-08-01 In diesem Band werden vielf ltige experimentelle Verfahren zur Bestimmung von L slichkeiten beschrieben Commission on Solubility Data der IUPAC beschlo die Qualit t der L slichkeitsdaten international anzugleichen und zu verbessern indem verl liche Verfahren zur Ermittlung dieser Daten zusammengefa t ver ffentlicht werden alle Datentypen wurden hier ber cksichtigt ausf hrliche Angaben zur Zuverl ssigkeit der einzelnen Verfahren und zur Auswahl der Aqueous Solution and the Phase Diagram Frederick Field Purdon, Victor Wallace Slater, 1946 jeweils geeigneten Methode

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