

Chemistry Unit 7 Review

701 – Inquiry Lab

1. Define...
 - a. Direct relationship: Two variables increase/decrease together
 - b. Indirect relationship: One variable increases, the other decreases
2. Identify whether the listed variables have a direct or indirect relationship to each other.
 - a. Moles and volume: Direct
 - b. Temperature and pressure: Direct
 - c. Pressure and volume: Indirect
 - d. Moles and pressure: Direct
3. Using the Ideal Gas Law equation, list 3 ways you could increase the volume of a balloon.
Increase the moles, increase the temperature, decrease the pressure (more so, the pressure drops as a result of greater volume)
4. Using the Ideal Gas Law equation, list 3 ways you could decrease the pressure within a syringe.
Decrease the moles, decrease the temperature, increase the volume
5. If the number of moles in a container is doubled, by how much should the pressure increase (assuming everything else is held constant)?
The pressure should also double

702 – Gas Laws Equation

1. What equation should you use when you are given just one set of variables?
 $PV = nRT$
2. You have a 4.5 L canister of O_2 that you store in a room that is $20\text{ }^{\circ}\text{C}$. The pressure of this tank is 13000 kPa. How many moles of O_2 are in this canister?
24 moles

Modeling Chemistry U7 Ws V2 0 Key

S Ben Porath



Modeling Chemistry U7 Ws V2 0 Key:

Modeling Chemistry U7 Ws V2 0 Key Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such may be the essence of the book **Modeling Chemistry U7 Ws V2 0 Key**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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