Deather

 An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

 $3.3 \text{ g} \times \underline{1 \text{ mole}}_{63.5 \text{ g}} = 5.2 \times 10^{4} \text{ mol or } 0.052 \text{ mol}$

Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?

4.42 g = 1 mole = 7.92 x 10° mol or 0.0792 mol 55.8 g

A raindrop has a mass of 0.050 g. How many moles of water does a raindrop contain?

0.050 g 1 mole = 2.78 x 10° mol → 2.8 x 10° mol

4. What mass of water would you need to have 15.0 moles of H2O?

15.0 moles 18.0 g = 2.70 x 10² g

 One box of Morton's Salt contains 737 grams. How many moles of sodium chloride is this?

737 g 1 mole = 12.6 mol

 A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). How many grams should the chef mass out?

0.050 moles 84.0 g = 4.2 g

7. Rust is iron(III) oxide. The owner of a 1959 Cadillac convertible wants to restore it by removing the rust with oxalic acid, but he needs to know how many moles of rust will be involved in the reaction. How many moles of iron(III) oxide are contained in 2 50 kg of rust?

Fe 111.6

48.0

2.50 kg 1000 q 1 mole = 15.7 mol Fe₂O₃ 1kg 159.6 g CI 35.5

58.5 a

No. 23.0

12.0

48.0

84.0

Modeling Chemistry Unit 1ws1

Peter Comba

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Chemistry Unit 1, 2nd Edition Bright Thinker, 2024-08 Practical Chemistry Unit 1,1987 **Chemical Modeling** Alan Hinchliffe, 1999-10-26 Chemical Modeling From Atoms to Liquids examines materials in terms of the basic properties of atoms molecules and polymer chains In particular the interactions between these fundamental building blocks and the intermolecular and intramolecular potentials are examined Fundamental theories of the constituent particles are covered introducing and developing classical mechanics and quantum mechanics from basics. These theories are then applied to modeling developing models from both classical mechanics and quantum mechanics. The book aims to make this subject both appealing and relevant whilst avoiding excessive mathematical rigour Carefully structured the text starts by introducing classical quantum and statistical mechanics before moving on to cover the modeling of solids gases and liquids To bring the subject alive many real life examples and applications have been included This book brings together the often scattered and diffuse background information that is essential for a full understanding of chemical modeling FEATURES Assumes no prior knowledge of modeling Brings together all the background information that a reader needs to know to fully understand modeling Develops classical and quantum mechanical theories from basics Avoids unnecessary mathematical rigour Includes a detailed mathematical Toolbox as a ready reference Includes brief descriptions and web addresses of key software packages CONTENTS Introduction Acknowledgements Software Packages Describing Macroscopic Systems Thermodynamics R sum of Classical Mechanics Modeling Simple Solids I Introduction to Quantum Mechanics Electric Multipoles Polarizabilities and Intermolecular Forces Some Statistical Ideas Applications of the Boltzmann Distribution Modeling Simple Solids II Molecular Mechanics Molecular Dynamics and Monte Carlo Techniques The Ideal Monatomic Gas Quantum Gases Introduction to Statistical Thermodynamics Modeling Atoms Diatomics Quantum Modeling of Larger Systems Describing Electron Correlation The Band Theory of Solids Modeling Polymeric Materials Modeling Liquids Appendices Suggestions for **Further Reading Index** Molecular Modelling for Beginners Alan Hinchliffe, 2011-08-17 A concise basic introduction to modelling and computational chemistry which focuses on the essentials including MM MC and MD along with a chapter devoted to QSAR and Discovery Chemistry Includes supporting website featuring background information full colour illustrations questions and answers tied into the text Visual Basic packages and many realistic examples with solutions Takes a hands on approach using state of the art software packages G03 W and or Hyperchem Gaussian gif files and sample outputs Revised with changes in emphasis and presentation to appeal to the modern student **Applying Molecular and Materials** Modeling Phillip Westmoreland, Peter A. Kollman, Anne M. Chaka, Peter T. Cummings, Keiji Morokuma, Matthew Neurock, Ellen B. Stechel, Priya Vashishta, 2002-10-31 Computational molecular and materials modeling has emerged to deliver solid technological impacts in the chemical pharmaceutical and materials industries It is not the all predictive science fiction that discouraged early adopters in the 1980s Rather it is proving a valuable aid to designing and developing new products and

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