

Assessment #3 Review

Name: _____

You must show all work for credit.

Due Date: _____ (No late packets will be accepted)

I. Balance Equations*Balance the equations*

- $\text{Mg} + \underline{2} \text{H}_2\text{O} \rightarrow \text{Mg}(\text{OH})_2 + \text{H}_2$
- $\text{Mg} + \underline{2} \text{NaBr} \rightarrow \text{MgBr}_2 + \underline{2} \text{Na}$
- $\text{C}_3\text{H}_8 + \underline{5} \text{O}_2 \rightarrow \underline{3} \text{CO}_2 + \underline{4} \text{H}_2\text{O}$
- $\underline{4} \text{P} + \underline{5} \text{O}_2 \rightarrow \underline{2} \text{P}_2\text{O}_5$
- $\text{Fe}_2(\text{SO}_4)_3 + \underline{6} \text{KOH} \rightarrow \underline{3} \text{K}_2\text{SO}_4 + \underline{2} \text{Fe}(\text{OH})_3$
- $\text{Fe}_2\text{O}_3 + \underline{3} \text{H}_2 \rightarrow \text{Fe} + \underline{3} \text{H}_2\text{O}$

7. How many moles of hydrogen gas react with 1 mole of Fe_2O_3 in reaction #6? 3

Write the balanced equation for problems # 8 and 9:

8. Hydrogen gas and nitrogen monoxide combine to form water and nitrogen gas



9. When sodium phosphate and calcium chloride are mixed they form calcium phosphate and sodium chloride



10. Using the following balanced equation, prove (or explain) the Law of Conservation of Mass.



2 N on left, 2 N on right, 6 H on left, 6 H on right, 2 on left, 2 on right.

11. Using the balanced equation shown in problem #10, explain what the coefficients represent.

2 N, 6 H, 2 on left, 2 on right.

12. Using the balanced equation shown in problem #10, list the products of the reaction.

2NH₃

13. Using the balanced equation shown in problem #10, list the reactants of the reaction.

N₂, 3H₂

II. Types of Reactions*Directions: Identify the type of reaction.*

S = synthesis

SD = single displacement

C = combustion

D = decomposition

DD = double displacement



Modeling Chemistry Unit 6 Test Review

LL Leslie



Modeling Chemistry Unit 6 Test Review:

This book delves into Modeling Chemistry Unit 6 Test Review. Modeling Chemistry Unit 6 Test Review is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Modeling Chemistry Unit 6 Test Review, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Modeling Chemistry Unit 6 Test Review
- Chapter 2: Essential Elements of Modeling Chemistry Unit 6 Test Review
- Chapter 3: Modeling Chemistry Unit 6 Test Review in Everyday Life
- Chapter 4: Modeling Chemistry Unit 6 Test Review in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Modeling Chemistry Unit 6 Test Review. This chapter will explore what Modeling Chemistry Unit 6 Test Review is, why Modeling Chemistry Unit 6 Test Review is vital, and how to effectively learn about Modeling Chemistry Unit 6 Test Review.
3. In chapter 2, the author will delve into the foundational concepts of Modeling Chemistry Unit 6 Test Review. This chapter will elucidate the essential principles that must be understood to grasp Modeling Chemistry Unit 6 Test Review in its entirety.
4. In chapter 3, the author will examine the practical applications of Modeling Chemistry Unit 6 Test Review in daily life. This chapter will showcase real-world examples of how Modeling Chemistry Unit 6 Test Review can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Modeling Chemistry Unit 6 Test Review in specific contexts. The fourth chapter will explore how Modeling Chemistry Unit 6 Test Review is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Modeling Chemistry Unit 6 Test Review. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Modeling Chemistry Unit 6 Test Review.

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