Chemistry Unit 3 Test Review

5.1 Mole and Avogadro's Constant

A particle is very small and can be:

- A single atom (Ca)
- A molecule with 2 or more atoms (molecular compound H2O)
- An ion with a positive or negative charge (CO₁2-)
- A formula unit with two or more ions (ionic compound NaCl)

The Mole

- · Chemists use the mole to represent a specific number of particles
- The number of particles it represents is called the Avogadro constant (6.02 × 10²⁹ particles/mol)
- The unit for mole is mol

Definitions

- Mole: One mole of any entity is 6.02 × 10²⁵ of the particles (atom, molecule, ion formula unit); the symbol is n; the unit is mol
- Avogadro's constant: The number of particles in one mole, 6.02 × 10²⁸; the symbol is Na; the unit is particles/mol
- 3. Particles:
- Atom or element (Eg. Zn)
- Molecule or covalent compound (Eg. H2O)
- lons or monatomic/polyatomic ions (Eg. Zn+2 and CO₁2-)
- Formula unit or ionic compound (Eg. LiCl)

5.2 Molar Mass

Basics: The Mass and The Mole

- The mass of carbon-12 is 12 atomic mass unit (12 u). Every other element has mass relative to one atom of carbon-12
- 1 mole of carbon-12 = 6.02 × 10²³ atoms of C-12 = 12g
- The average atomic mass of any element, in atomic mass units, is numerically
 equivalent to the mass of one mole of atoms (6.02 × 10²⁸ atoms) of that element

Motar Mass: The mass of one mole of its particle – atoms, molecules, ions, or formula units, in grams per mole (g/mol)

6.1 Percentage Composition

Law of Definite Proportions:

- · Elements in a chemical compound are always present in the same proportions by mass
- Law of constant composition: all molecules or formula units of a compound will have the same chemical composition

Modeling Chemistry Unit 3 Test Answers

Rachel S Tattersall

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web the weary blues is a poem about the pain and suffering conveyed through the blues and how african americans used this musical genre as an outlet to express their pain and struggle by implementing elements of the blues throughout the poem hughes adds a strong musical and cultural element to help readers connect to it

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