

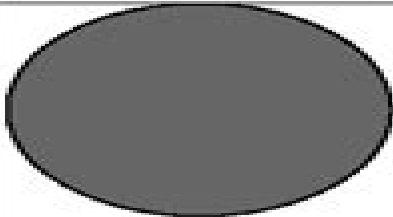
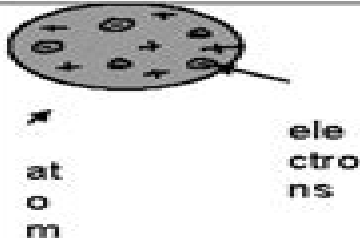

# Atomic Theory Worksheet

*Objective: You will create an abbreviated atomic theory timeline to show how the scientists involved in the atomic theory worked together to construct the modern model of the atom.*

## Part 1 – Atomic Theory Timeline

Choose 3 scientists from the options below and complete the table.

Options: John Dalton, Niels Bohr, JJ Thomson, Erwin Schrödinger, Ernest Rutherford

| Content   | Scientist 1   | Scientist 2   | Scientist 3  |
|---|---|---|--|
| Timeframe [in order]  | 1803  | 1897  | 1911   |
| Scientist Name  | John Dalton   | J.J. Thompson   | Ernest Rutherford  |
| What did he contribute to the atomic theory?  | He contributed that atoms can not be made or destroyed, a chemical reaction only can happen when atoms are rearranged, as well all atoms of a specific element are the same in mass and properties. | He contributed that atoms have charged particles that are called electrons, he also thought that matter is electronically neutral, he also thought that electrons are dispersed evenly through an atom. | He contributed that an atom has a very small positively charged center called the nucleus and it contains most of the atoms mass. He also added that atoms have a cloud of electrons that circle around the nucleus. |
| Draw a sketch of their model [by hand or using the drawing tool] and label all parts. |    | <br>atom<br>electrons   | <br>electrons<br>nucleus  |

## Part 2 – Reflection Questions

1. Describe two details of the modern atomic model.

Atoms have a nucleus of protons and neutrons and a cloud around the nucleus that contains the electrons.

2. Explain how the atomic model has changed over time.

The Rutherford model got replaced by the Bohr model because scientists found out that electrons have a specific orbit around the nucleus.

# Modern Atomic Theory Answers

**J Elliott**



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## **Table of Contents Modern Atomic Theory Answers**

1. Understanding the eBook Modern Atomic Theory Answers
  - The Rise of Digital Reading Modern Atomic Theory Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Modern Atomic Theory Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Modern Atomic Theory Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Atomic Theory Answers
  - Personalized Recommendations
  - Modern Atomic Theory Answers User Reviews and Ratings
  - Modern Atomic Theory Answers and Bestseller Lists
5. Accessing Modern Atomic Theory Answers Free and Paid eBooks
  - Modern Atomic Theory Answers Public Domain eBooks
  - Modern Atomic Theory Answers eBook Subscription Services
  - Modern Atomic Theory Answers Budget-Friendly Options

6. Navigating Modern Atomic Theory Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Modern Atomic Theory Answers Compatibility with Devices
  - Modern Atomic Theory Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Modern Atomic Theory Answers
  - Highlighting and Note-Taking Modern Atomic Theory Answers
  - Interactive Elements Modern Atomic Theory Answers
8. Staying Engaged with Modern Atomic Theory Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Modern Atomic Theory Answers
9. Balancing eBooks and Physical Books Modern Atomic Theory Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Modern Atomic Theory Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Modern Atomic Theory Answers
  - Setting Reading Goals Modern Atomic Theory Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Atomic Theory Answers
  - Fact-Checking eBook Content of Modern Atomic Theory Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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