# **Atomic Theory Worksheet**

Objective: You will create an abbreviated atomic theory timeline to show how the scientists involved in the atomic theory worked together to construct the modern model of the atom.

#### Part 1 - Atomic Theory Timeline

Choose 3 scientists from the options below and complete the table.

Options: John Dalton, Niels Bohr, JJ Thomson, Envin Schrödinger, Ernest Rutherford

Content	Scientist 1	Scientist 2	Scientist 3
Timeframe [in order]	1803	1897	1911
Scientist Name	John Dalton	J.J. Thompson	Emest Rutherford
What did he contribute to the atomic theory?	He contributed that atoms can not be made or destroyed, a chemical reaction only can happen when atoms are rearranged, as well all atoms of a specific element are the same in mass and properties.	He contributed that atoms have charged particles that are called electrons, he also thought that matter is electronically neutral, he also thought that electrons are dispersed evenly through an atom.	He contributed that an atom has a very small positively charged center called the nucleus and it contains most of the atoms mass. He also added that atoms have a cloud of electrons that circle around the nucleus.
Draw a sketch of their model [by hand or using the drawing toof] and label all parts.		ele ctro ns m	ele n ctro u ns c

#### Part 2 - Reflection Questions

Describe two details of the modern atomic model.

Atoms have a nucleus of protons and neutrons and a cloud around the nucleus that contains the electrons.

2. Explain how the atomic model has changed over time.

The Rutherford model got replaced by the bohr model because scientists found out that electrons have a specific orbit around the nucleus.

# **Modern Atomic Theory Concept Review Answers**

Morris Hein, Susan Arena, Cary Willard

#### **Modern Atomic Theory Concept Review Answers:**

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Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

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