



MODERN AUTOMOTIVE TECHNOLOGY

by James E. Duffy, ©2014, 8th Edition

Suggested Grade Levels 9-12

Modern Automotive Technology details the construction, operation, diagnosis, service, and repair of late-model automobiles and light trucks. This comprehensive textbook uses a building-block approach that starts with the fundamental principles of system operation and progresses gradually to complex diagnostic and service procedures. Short sentences, concise definitions, and thousands of color illustrations help students learn quickly and easily. The new edition of Modern Automotive Technology provides coverage of the latest developments in the automotive field and is correlated to the 2012 NATEF Task List, A new Fundamentals of Electricity and Electronics section provides students with the background needed to troubleshoot and repair the complex electrical/electronic systems found on today's vehicles. Updated information on hybrid drive systems has been integrated throughout the textbook, and a new hybrid drive system service chapter details the diagnosis and repair of these important systems.

Organized around the ASE automobile test areas, Modern Automotive Technology is a valuable resource for students preparing for a career in automotive technology, as well as experienced technicians preparing for the ASE certification tests.

www.g-w.com/modern-automotive-technology-2014

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2ag\ aq\ cr\ 2o\ 7\ 2\ aq\ rightarrow\ ag\ 2cr\ 2o\ 7\ s\ label\ 4\ 2\ 3$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water h aq oh aq h 2 o l the spectator ions in koh aq and hno 3 aq are k aq and no 3 aq how to identify and list the

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web the ag aq and cl aq ions become agcl s but the na aq ions and the no 3 aq ions stay as na aq ions and no 3 aq ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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