

Homework: Week 4 homework

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HW Score: 0%, 0 of 1 pt

CC 12-1 (algorithmic)

Question Help



The graph shows the ideal gas law relationship ($PV = nRT$) between pressure (P) and temperature (T).

Click the icon to view the graph of the ideal gas law relationship.

Click the icon to view the value and units of gas constant R for Ideal Gas Law.

(a) What are the units of the slope (0.0087)?

The units of the slope are $\frac{\text{atm}}{\text{K}}$.

(b) If the tank has a volume of 24 liters and is filled with nitrogen (formula, N_2 ; molecular weight, 28 grams per mole), what is the mass of gas in the tank (m) in units of grams?

The mass is grams. (Round your answer to three significant digits.)

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$$R = 8.314 \frac{\text{Pa} \cdot \text{m}^3}{\text{mol} \cdot \text{K}}$$
$$= 0.08206 \frac{\text{atm} \cdot \text{L}}{\text{mol} \cdot \text{K}}$$

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