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Chemistry – Unit 6 Worksheet 2

Balance the following equations by inserting the proper coefficients.

1. ____ SO_2 + ____ O_2 $\rightarrow$ ____ SO_3
2. ____ CH_4 + ____ O_2 $\rightarrow$ ____ CO + ____ H_2O
3. ____ P + ____ Cl_2 $\rightarrow$ ____ PCl_3
4. ____ CO + ____ O_2 $\rightarrow$ ____ CO_2
5. ____ CH_4 + ____ O_2 $\rightarrow$ ____ CH_3OH
6. ____ Li + ____ Br_2 $\rightarrow$ ____ LiBr
7. ____ Al_2O_3 $\rightarrow$ ____ Al + ____ O_2
8. ____ Na + ____ H_2O $\rightarrow$ ____ NaOH + ____ H_2
9. ____ CO_2 + ____ H_2O $\rightarrow$ ____ $\text{C}_6\text{H}_{12}\text{O}_6$ + ____ O_2
10. ____ H_2SO_4 + ____ NaCl $\rightarrow$ ____ HCl + ____ Na_2SO_4

Two reactions used to get rid of sulfur dioxide, a pollutant from burning coal:

11. ____ H_2 + ____ SO_2 $\rightarrow$ ____ H_2S + ____ H_2O
12. ____ CaCO_3 + ____ SO_2 + ____ O_2 $\rightarrow$ ____ CaSO_4 + ____ CO_2
13. ____ AgNO_3 + ____ CaCl_2 $\rightarrow$ ____ AgCl + ____ $\text{Ca}(\text{NO}_3)_2$
14. ____ HCl + ____ $\text{Ba}(\text{OH})_2$ $\rightarrow$ ____ BaCl_2 + ____ H_2O
15. ____ H_3PO_4 + ____ NaOH $\rightarrow$ ____ Na_3PO_4 + ____ H_2O
16. ____ $\text{Pb}(\text{NO}_3)_2$ + ____ KI $\rightarrow$ ____ PbI_2 + ____ KNO_3
17. ____ CuO + ____ NH_3 $\rightarrow$ ____ N_2 + ____ Cu + ____ H_2O
18. ____ $\text{C}_2\text{H}_5\text{OH}$ + ____ O_2 $\rightarrow$ ____ CO_2 + ____ H_2O
19. ____ C_2H_6 + ____ O_2 $\rightarrow$ ____ CH_3COOH + ____ H_2O
20. ____ NO_2 + ____ H_2O $\rightarrow$ ____ HNO_3 + ____ NO

Write the formulas of the reactants and products - including the symbols for the state - (s), (l), (g), (aq) - then balance the equations.

1. Gaseous ammonia (NH_3) reacts with hydrogen chloride gas to form solid ammonium chloride.

Modeling Chemistry U5 Ws1 V1 1 Key

Michael Hoelscher



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