Chemistry MCQ (UNIT 4)

Q1 Select the incorrect statement from the following option.

 a) Permanent hardness is due to dissolved chlorides and sulphates of calcium and magnesium

b) It can be removed by mere boiling of water

c) It is also known as non-alkaline hardness

d) The difference between the total hardness and the alkaline hardness gives the non-alkaline hardness

Ans: (b)

Q2 Select the incorrect statement from the following option.

a) The taste of hard water is better than soft water

b) The dissolved calcium in hard water can help to produce strong teeth

 c) Hard water coats the lead piping with a layer of insoluble calcium carbonate which prevents poisonous lead dissolving in water

d) Boiler feed water should also be hard in nature

Ans: (d)

Q3. Hardness of water is conventionally expressed in terms of equivalent amount of

a) H.CO.

b) MgCO.

c) CaCO.

d) Na. CO,

Ans: (c)

Q 4. Which of the following is not a unit of hardness?

a) Parts per million

b) Degree centigrade

c) Degree clarke

d) Degree French

Ans: (b)

Q 5. How many grams of MgCO, dissolved per litre gives 84 ppm hardness?

a) 70.56 mg/L

b) 48.23 mg/L

c) 81.49mg/L

d) 66.12 mg/L

Ans: (a)

Modeling Chemistry Unit 4 Describing Substances Objectives Answers

S Ben Porath

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