

## Chemistry – Unit 5 Worksheet 2

1. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

$$3.3 \text{ g} \times \frac{1 \text{ mole}}{63.5 \text{ g}} = 5.2 \times 10^{-2} \text{ mol or } 0.052 \text{ mol}$$

2. Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?

$$4.42 \text{ g} \times \frac{1 \text{ mole}}{55.8 \text{ g}} = 7.92 \times 10^{-2} \text{ mol or } 0.0792 \text{ mol}$$

3. A raindrop has a mass of 0.050 g. How many moles of water does a raindrop contain?

$$0.050 \text{ g} \times \frac{1 \text{ mole}}{18.0 \text{ g}} = 2.78 \times 10^{-3} \text{ mol} \rightarrow 2.8 \times 10^{-3} \text{ mol}$$

4. What mass of water would you need to have 15.0 moles of H<sub>2</sub>O?

$$15.0 \text{ moles} \times \frac{18.0 \text{ g}}{1 \text{ mole}} = 2.70 \times 10^2 \text{ g}$$

5. One box of Morton's Salt contains 737 grams. How many moles of sodium chloride is this?

$$737 \text{ g} \times \frac{1 \text{ mole}}{58.5 \text{ g}} = 12.6 \text{ mol}$$

Na	23.0
Cl	35.5
	<hr/> 58.5 g

6. A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). How many grams should the chef mass out?

$$0.050 \text{ moles} \times \frac{84.0 \text{ g}}{1 \text{ mole}} = 4.2 \text{ g}$$

Na	23.0
H	1.0
C	12.0
O	48.0
	<hr/> 84.0

7. Rust is iron(III) oxide. The owner of a 1959 Cadillac convertible wants to restore it by removing the rust with oxalic acid, but he needs to know how many moles of rust will be involved in the reaction. How many moles of iron(III) oxide are contained in 2.50 kg of rust?

$$2.50 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ mole}}{159.6 \text{ g}} = 15.7 \text{ mol Fe}_2\text{O}_3$$

Fe	111.6
O	48.0
	<hr/> 159.6

# Modeling Chemistry Unit 5

**Hongru Du**



## Modeling Chemistry Unit 5:

**Chemistry in the Community** American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design *Resources in Education* ,1980 **Computer Aided Design of Polymers and Composites** D. H. Kaelble, 1985-02-01 *Advancements of Mass Spectrometry in Biomedical Research* Alisa G. Woods, Costel C. Darie, 2019-07-25 This volume explores the use of mass spectrometry for biomedical applications Chapters focus on specific therapeutic areas such as oncology infectious disease and psychiatry Additional chapters focus on methodology technologies and instrumentation as well as on analysis of protein protein interactions protein quantitation and protein post translational modifications Various omics fields such as proteomics metabolomics glycomics lipidomics and adductomics are also covered Applications of mass spectrometry in biotechnological and pharmaceutical industry are also discussed This volume provides readers with a comprehensive and informative manual that will allow them to appreciate mass spectrometry and proteomic research but also to initiate and improve their own work This book acts as a technical guide as well as a conceptual guide to the newest information in this exciting field *Federal Register* ,2012-03 *Journal of Applied Chemistry of the USSR* ,1988 **The American Curriculum** George Willis, 1993 At a time marked by strong demands for educational reform the American school curriculum is a topic of special concern This volume provides a comprehensive historical record of the evolution of the curriculum in America from the colonial period to the present day The editors have compiled a collection of influential and representative documents in primary secondary and higher education in the United States Each document is introduced by a short essay that discusses its historical context and significance The result is a valuable chronicle of the development of the American school curriculum The work begins with an introductory piece that overviews the development of the curriculum and surveys the most important works on curriculum history The introduction is followed by excerpts from 34 documents representative of the school curriculum from The Rules and Course of Study of Harvard College 1642 to the 1983 report A Nation at Risk The Imperative for Educational Reform The essay that introduces each document closes with a brief bibliography and the volume concludes with a more extensive list of sources for further reading By consulting this reference historians and educators can trace the development of the curriculum over the last 350 years *Paperbound Books in Print* ,1992 **The Journal of the Assembly, During the ... Session of the Legislature of the State of California** California. Legislature. Assembly, 1907 *Earth Science* ,2001 **General Catalogue** Carnegie Institute of Technology, 1915 *Ceramic Abstracts* American Ceramic Society, 1926 **Prentice Hall Science Explorer** Michael J. Padilla, 2002 **Pearson's Magazine** ,1911 Vol 49 no 9 Sept 1922 accompanied by a separately paged section entitled ERA electronic reactions of Abrams **Construction and Use of Atomic and Molecular Models** H. Bassow, 1968 **Manufacturers Record** ,1928 **Loyola Educational Index** Austin Guildford Schmidt, 1928

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