

Modelling Chemistry Second Semester Exam Review 2017/18

Unit 5 – Counting Particles Too Small to See

Learning Targets-I CAN:

1. Use a weighed sample of a substance and experimental data to determine the relative mass of a substance.
2. Use experimental data to determine the relative mass of anionic or cationic radicals without actually counting them.
3. Use experimental data to determine the relative mass of a molecule.
4. Show the chemical formulae and valencies, representing the relative mass.
5. Show the relative molecular mass of:
 - a. the reactant substances within a sample
 - b. the product substance or substances in the sample
6. Show the relative of moles of a substance and:
 - a. the mass of the sample
 - b. the relative of masses of reactants or products in the sample
7. Show the formulae of compounds, determine the % composition.
8. Show the percentage composition of a sample, determine the empirical formula of the compound.
9. Show the empirical formula, determine the relative mass of the compound, determine the molecular formula.

Modeling Chemistry Semester Final Review

James Millikin University



Modeling Chemistry Semester Final Review:

Catalogue Number. Course Catalog Anonymous, 2025-08-07 Reprint of the original first published in 1876 The Antigonos publishing house specialises in the publication of reprints of historical books We make sure that these works are made available to the public in good condition in order to preserve their cultural heritage **Using Physical Models of Biomolecules to Teach Concepts of Biochemical Structure in Introductory Undergraduate Chemistry** John Yi, 2004

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House documents, 1889 **Announcements for Register of Alumni** University of Michigan. College of Pharmacy, 1893 *Northwest Journal of Education*, 1913 *Technical Book Review Index*, 1979 **Mathematical Reviews**, 2008 **Bulletin**, 1927 *Medical Review of Reviews*, 1930 Index medicus in v 1 30 1895 1924 *The Temple University Catalogue* Temple University, 1917

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