

Unit 3 - Worksheet 1 – E Bar Charts

- A. For each of the situations described below, draw a Temperature vs. time (energy) graph for the situation.
 B. Then, use an energy bar chart to represent the ways that energy is stored in the system and flows into or out of the system.
 C. Below each diagram, briefly describe how the **motion** and/or **arrangement** of the molecules **changes** from the initial to the final state.

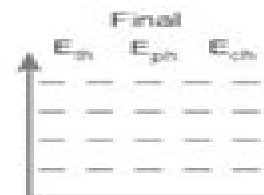
1. A cup of hot coffee cools as it sits on the table.

A. Temp/time Graph:

B. Bar Chart:



Energy Flow



C. Molecule Description:

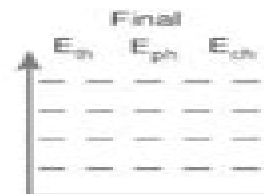
2. A can of cold soda warms as it is left on the counter.

A. Temp/time Graph:

B. Bar Chart:



Energy Flow



C. Molecule Description:

3. A tray of water (20 °C) is placed in the freezer and turns into ice cubes (- 8 °C)

A. Temp/time Graph:

B. Bar Chart:



Energy Flow



C. Molecule Description:

Modeling Chemistry U3 Ws 3 V2

R Barnett



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