

Chemistry – Unit 6 Notes 1 - Thomson Model of the Atom

J. J. Thomson performed experiments with cathode rays in an attempt to understand electricity – which was still a mystery in the late 1800s. Use the website *A Look Inside the Atom*¹ to find the conclusions that Thomson and other physicists drew regarding the mysterious cathode rays.

Thomson's 1897 Experiments - state the conclusions Thomson drew from each of his famous cathode ray experiments:

1. **First Experiment:** Thomson directed the beam at an electrometer and tried to separate the evidence of charge from the path of the beam. *What connection did Thomson find between charge and the cathode rays? Was the charge positive or negative?*
2. **Second Experiment:** Thomson tried passing the cathode ray through an electric field. *How did cathode ray beam behave when it passed through an electric field? What did he conclude after his second experiment?*
3. **Third Experiment:** Thomson did some careful measurements on how much the path of the cathode ray was bent in a magnetic field and how much energy they carried. From this work Thomson could describe the mass/charge ratio of the cathode ray particles. *What amazing result did Thomson find?*

Thomson's Atomic Model: Thomson presented three hypotheses from his experiments. Only two were accepted by physicists – in fact the third was shown to be wrong! From the first two came a model of the atom known as the *Plum Pudding* model. Complete the atom drawing below to illustrate Thomson's plum pudding model. Explain how this fits with his observations.



¹ <https://www.aip.org/history/exhibits/electron/jjrays.htm>
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RJ Alexander



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Embracing the Tune of Phrase: An Psychological Symphony within **Modeling Chemistry U6 Ws 4 V0**

In some sort of used by screens and the ceaseless chatter of instant conversation, the melodic elegance and psychological symphony produced by the published word frequently disappear into the background, eclipsed by the constant noise and interruptions that permeate our lives. However, nestled within the pages of **Modeling Chemistry U6 Ws 4 V0** an enchanting literary prize full of raw thoughts, lies an immersive symphony waiting to be embraced. Crafted by a wonderful musician of language, that captivating masterpiece conducts visitors on a psychological trip, well unraveling the concealed melodies and profound impact resonating within each cautiously constructed phrase. Within the depths of this emotional assessment, we shall discover the book is main harmonies, analyze their enthralling writing type, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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