Chemistry Unit 8 Worksheet 4 Samples of Every Kind of Problem

On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedure outlined in class. Be sure to circle your final answer.

 Calculate the number of moles of potassium chlorate, KClO₃ (s), that must decompose to produce potassium chloride, KCl (s), and 1.8 moles of oxygen gas.

Equation:	2 KClO ₃ +	≥ 2 KCl +	3 02
Before:	1.2		0
Change:	-1.2	+1.2	+1.8
Afters	.0	1.2	1.8

1.2 moles KCRO, needed

2. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 2.43 g of magnesium?

Equations	Non-	2 HCl >	MgCl ₂ +	882
Before:	0.100	7		0
Change:	-0.100	-0.200 +	0.100 +	0.100
Afters		2	0.100	0.100

0.200 protes HCI needed

0.73 g x 1 mole = .0040 mole 183 gr

 Ethane, C₂H₆ reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.20 moles of ethane?

Equation:	2 C2H6	+ 7 O ₂	→ 4 CO₂	+ 6 H ₂ O
Before:	2.20	5.5	0	0.0
Change:	-2.20	~ 7.70	+ 4,40	+ 6,60
After:	0	3.5	4.40	6.60

O2 needed: 7,70 mole x 32.0 g = 246 g O2 required I mode

 Determine the mass of sodium nitrate produced when 0.73 g of nickel (II) nitrate reacts with sodium hydroxide according to the following equation: Ni(NO3)2 + 2 NaOH --> Ni(OH)2 + 2 NaNO3

> Ni(NO₃)₂ + 2 NaOH -> Ni(OH)₂ + 2 NaNO₃ Equation: Before: 0.0040 No. 26 -0.0040- O. GOOSIO **-0.0040** +0.06900

N. S. 0.0080 Afflors 0.0640 NaNO₂ produced: 0.0080 mole x 85.0 g = 0.68 g 1 mode

Change:

Modeling Chemistry Unit 13 Packet

Anthony S. Fauci

Modeling Chemistry Unit 13 Packet:

Modeling Chemistry Unit 13 Packet Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Modeling Chemistry Unit 13 Packet**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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