

7 • Molecular Structure

STATION 3 • POLARITY

Determine the shape of the molecule and decide whether each compound is polar or nonpolar.

PH_3 $3 \times 1 = 3$ $\text{H} : \underset{\cdot\cdot}{\underset{\cdot\cdot}{\text{P}}} : \text{H}$ Polar	CF_4 $4 \times 4 = 16$ $\begin{array}{c} :\ddot{\text{F}}: \\ :\ddot{\text{F}}: \text{C} : \ddot{\text{F}}: \\ :\ddot{\text{F}}: \end{array}$ non-polar	BH_3 $3 \times 3 = 6$ $\text{H} : \underset{\cdot\cdot}{\underset{\cdot\cdot}{\text{B}}} : \text{H}$ non-polar
SO_2 6 $\begin{array}{c} :\ddot{\text{O}}: \text{S} : \ddot{\text{O}}: \\ \text{or } :\ddot{\text{O}}: \text{S} : \ddot{\text{O}}: \\ \text{or } :\ddot{\text{O}}: \text{S} : \ddot{\text{O}}: \end{array}$ Polar	CO_2 $12 \times 4 = 16$ $\begin{array}{c} :\ddot{\text{O}}: \text{C} : \ddot{\text{O}}: \\ \text{or } :\ddot{\text{O}}: \text{C} : \ddot{\text{O}}: \end{array}$ non-polar	SCl_2 $6 + 6 = 12$ $\begin{array}{c} :\ddot{\text{Cl}}: \text{S} : \ddot{\text{Cl}}: \\ \text{or } :\ddot{\text{Cl}}: \text{S} : \ddot{\text{Cl}}: \end{array}$ polar

7 • Molecular Structure

STATION 4 • IMF's AND ATTRACTIONS

Match the statement with the Inter-Particle Forces

A. Dipole-Dipole Interactions

B. Covalent bonds

C. Hydrogen-bonding

D. Ionic Bonds

E. Metallic Bonds

F. London Dispersion Forces

- F 1. Is used to explain why BP of $\text{Xe} > \text{Kr} > \text{Ar} > \text{Ne} > \text{He}$
- B 2. Is present in ~~Complete~~ all ~~Compounds~~ Compounds
- E 3. Is used to explain why Cu is a good conductor
- D 4. Is used to explain why $\text{NaCl}(\text{l})$ is a good conductor
- A 5. Is used to explain why ICI has a higher BP than Br_2
- C 6. Is used to explain why boiling point of CH_3OH is higher than CH_4
- E 7. A lattice of positive ions in a sea of delocalized electrons
- A 8. Positive ends of molecules attract negative ends of other molecules
- E 9. Lattice of alternating positively and negatively charged particles
- C 10. Hydrogen atoms tethered between small, highly electronegative atoms N, O, and F.
- F 11. Polarized electron clouds induce dipoles in their neighboring atoms

Modeling Chemistry Unit 7 Review Answers

M Woodhall



Modeling Chemistry Unit 7 Review Answers:

Commerce Business Daily ,2001-03 **Engineering Mechanics: Statics** Sheri D. Sheppard,Thalia Anagnos,Sarah L. Billington,2017-06-14 Mechanics courses tend to provide engineering students with a precise mathematical but less than engaging experience Students often view the traditional approach as a mysterious body of facts and tricks that allow idealized cases to be solved When confronted with more realistic systems they are often at a loss as to how to proceed To address this issue this course empowers students to tackle meaningful problems at an early stage in their studies

Engineering Mechanics Statics First Edition begins with a readable overview of the concepts of mechanics Important equations are introduced but the emphasis is on developing a feel for forces and moments and for how loads are transferred through structures and machines From that foundation the course helps lay a motivational framework for students to build their skills in solving engineering problems *Encyclopedia of Renewable and Sustainable Materials* ,2020-01-09

Encyclopedia of Renewable and Sustainable Materials Five Volume Set provides a comprehensive overview covering research and development on all aspects of renewable recyclable and sustainable materials The use of renewable and sustainable materials in building construction the automotive sector energy textiles and others can create markets for agricultural products and additional revenue streams for farmers as well as significantly reduce carbon dioxide CO2 emissions manufacturing energy requirements manufacturing costs and waste This book provides researchers students and professionals in materials science and engineering with tactics and information as they face increasingly complex challenges around the development selection and use of construction and manufacturing materials Covers a broad range of topics not available elsewhere in one resource Arranged thematically for ease of navigation Discusses key features on processing use application and the environmental benefits of renewable and sustainable materials Contains a special focus on sustainability that will lead to the reduction of carbon emissions and enhance protection of the natural environment with regard to sustainable materials **Books in Print** ,1994 **Cumulated Index Medicus** ,2000 Chemistry and Physics of Coal Utilization-1980 Bernard R. Cooper,Leonidas Petrakis,1981 *Energy Research Abstracts* ,1982 **Monthly Catalog of United States Government Publications** ,1996 The Environment Index ,1986 **The United States Catalog** Mary Burnham,Carol Hurd,1928 **Research Centers Directory** ,1960 Research institutes foundations centers bureaus laboratories experiment stations and other similar nonprofit facilities organizations and activities in the United States and Canada Entry gives identifying and descriptive information of staff and work Institutional research centers and subject indexes 5th ed 5491 entries 6th ed 6268 entries **Spring Meeting** American Geophysical Union. Meeting,1998

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