

Chemistry Unit 3 Worksheet 4

Examples of Every Kind of Problem

This is a separate sheet of paper, so this is a complete solution to each of the problems. You can follow the problems without an answer. The answer is under each kind of problem.

1. Calculate the reaction of sulfur of potassium chlorate, KClO_3 , and potassium chlorate to produce potassium chloride KCl and O_2 reaction of oxygen gas.



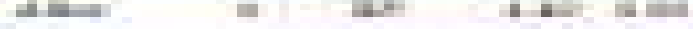
1. 174 g of KClO_3 is needed

2. As a single displacement reaction, potassium metal reacts with hydrochloric acid to produce potassium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 5.0 g of potassium?



2. 0.067 moles of HCl is needed

3. Ethane, C_2H_6 , reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 10.0 g of ethane?



3. 336 g of O_2 is needed to react with 10.0 g of C_2H_6

4. Sodium metal reacts with sulfur dioxide to produce sodium sulfide and sodium sulfate. How many moles of sodium metal are needed to react with 10.0 g of sulfur dioxide?



4. 0.5 moles of Na is needed to react with 10.0 g of SO_2

Modeling Chemistry Unit 8 4

Sasha Hu



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Modeling Chemistry Unit 8 4 Book Review: Unveiling the Power of Words

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