

CHM 2210 Lab: Melting Point

Introduction

The melting point of a solid is the temperature at which the solid melts (changes to a liquid) at 1 atm of pressure. Melting point is determined by slow heating of a small solid sample placed in a capillary tube. The melting point should be reported as a melting point range: from the temperature at which the first droplet of liquid is observed to the temperature at which the sample becomes a clear liquid, for example: 96.7 – 98.9°C.

A pure solid (pure enough for most purposes) usually has a sharp melting point which means that it melts within a range of 2°C or less.

An impure solid melts at a broader range (3°C or more) and at depressed temperature.

Using melting point to find the identity of a solid is not sufficient because there are too many substances with the same melting points.

However, you can use a mixed melting point as a tool for identification of a solid.

If you mix two different solids with the same melting point, the melting point of the mixture will be broad and depressed. The melting point of a solid and a mixture of this solid with the authentic solid should be the same.

Example:

Measured/Observed melting point of an unknown solid X: 128.0 – 129.5°C

Melting point of known solid A: 130°C

Melting point of known solid B: 158°C

Measured melting point of mixture X + A: 127.5 – 129.0°C

Measured melting point of mixture X + B: 133.5 – 142.0°C

Identity of unknown X: solid A

In the future you will measure melting points of solids mostly to determine their purity.

- **Preparation of Sample**

A solid sample should be dried and crushed with a metal spatula or bottom of a test tube.

If a solid is used for mixed melting point, the 1:1 mixture should be crushed together into a homogeneous mixture.

The pulverized solid should be mounded up and a small (1 – 2 mm) plug of sample should be pushed into the open end of the capillary tube. The sample should be knocked down to the sealed end of the capillary tube by tapping its sealed end on the surface or dropping it into a long glass tubing.

The sample should be 1-2 mm long and packed firmly.

If the sample is too large, the measured melting point may be too high.

If the sample is loosely packed, it may be difficult to determine the melting point range because of the air present.

Melting Point Lab For High School

**Mary Margaret Capraro, Robert
Capraro**



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Fostering Sustained Learning Among Undergraduate Students: Emerging Research and Opportunities Chemosit, Caroline, Rugutt, John, Rugutt, Joseph K., 2017-03-03 Keeping students engaged and receptive to learning can at times be a challenge. However, by the implementation of new methods and pedagogies, instructors can strengthen the drive to learn among their students. *Fostering Sustained Learning Among Undergraduate Students: Emerging Research and Opportunities* is an essential publication for the latest scholarly information on methods to inculcate student learning with a focus on implications to institutional policy and practices. Featuring coverage on topics such as financial aid, student motivation, and mentorship, this book is ideally designed for academicians, practitioners, and researchers seeking novel perspectives on the learning process and instruction methods.

Gourmet Lab Sarah Reeves Young, 2011 Hands on inquiry based and relevant to every student. OCOs life Gourmet Lab serves up a full menu of activities for science teachers of grades 6-12. This collection of 15 hands on experiments, each of which includes a full set of both student and teacher pages, challenges students to take on the role of scientist and chef as they boil, bake, and toast their way to a better understanding of science concepts from chemistry, biology, and physics. By cooking edible items such as pancakes and butterscotch, students have the opportunity to learn about physical changes in states of matter, acids and bases, biochemistry, and molecular structure. The Teacher pages include Standards addressed in each lab, a vocabulary list, safety protocols, materials required, procedures, data analysis, student questions, answer key, and conclusions, and connections to spur wrap up class discussions. Cross curricular notes are also included to highlight the lesson's connection to subjects such as math and literacy. Finally, optional extensions for both middle school and high school levels detail how to explore each concept further. What better topic than food to engage students to explore science in the natural world.

Catalogue of the Ann Arbor High School for the Academic Year Ann Arbor High School (Mich.), 1903

Chemistry Dorin, Prentice Hall Dictionary Editors, 1992

Michigan School Moderator, 1908

A Companion To Interdisciplinary Stem Project-Based Learning Mary Margaret Capraro, Robert Capraro, 2016-07-18 This text contains 25 Project Based Learning (PBL) lessons written by a combination of undergraduate preservice teachers, inservice teachers, and graduate students. Everyone who wrote a chapter strives to improve STEM education to help others implement standards based STEM instruction that takes learning in isolation to greater accountability through integrated and meaningful tasks that answer the question every teacher dreads: When am I going to use this? The PBLs were written to implement in middle and high school classrooms. All of them are interdisciplinary in nature. We have divided them into six themes: construction and design, water environment, mixtures, technology, nutrition, and genetics. Each lesson contains a schedule at a glance and the well defined outcome so you can quickly see how a particular PBL fits into your curriculum. Objectives are listed along with STEM connections written as objectives. We have included all materials needed and then each day of activities including an imbedded engagement.

exploration explanation evaluation including rubrics and extension We have tried to include everything necessary for successful implementation This practical book is the perfect companion to the handbook for learning about implementing PBLs Project Based Learning An Integrated Science Technology Engineering and Mathematics STEM Approach second edition **Catalogue** Ann Arbor High School (Mich.),1903 **Managing Technological Innovation** Frederick Betz,2003-07-03 Technology management as a field came together during the 1980s in response to the question of how society could deliberately create new technology and exploit it in economic development This updated edition introduces technology management covers the importance of managing information technologies and compares them to existing physical technologies **Cooperative Chemistry Lab Manual** Cooper,2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations **Chemical Abstracts** ,1927 **Teaching Science Online** Dietmar Kennepohl,2023-07-03 With the increasing focus on science education growing attention is being paid to how science is taught Educators in science and science related disciplines are recognizing that distance delivery opens up new opportunities for delivering information providing interactivity collaborative opportunities and feedback as well as for increasing access for students This book presents the guidance of expert science educators from the US and from around the globe They describe key concepts delivery modes and emerging technologies and offer models of practice The book places particular emphasis on experimentation lab and field work as they are fundamentally part of the education in most scientific disciplines Chapters include Discipline methodology and teaching strategies in the specific areas of physics biology chemistry and earth sciences An overview of the important and appropriate learning technologies ICTs for each major science Best practices for establishing and maintaining a successful course online Insights and tips for handling practical components like laboratories and field work Coverage of breaking topics including MOOCs learning analytics open educational resources and m learning Strategies for engaging your students online *Moderator-topics* ,1907 **The Organic Chem Lab Survival Manual** James W. Zubrick,1988-01-18 This clearly written presentation of basic organic chemistry techniques is designed to quickly build student proficiency in a wide range of experimental procedures Written in an informal style and supported by plentiful illustrations this manual fully explains the reasons for using a particular sequence of operations and offers numerous examples of what can go wrong in each experiment In addition it shows how to manipulate and support the apparatus and substances Throughout emphasis is on correct sample preparation and instrument operation as well as laboratory safety All procedures including complicated operations such as distillation and extraction are treated in a straightforward fashion Illustrated World ... ,1906 **Directory, Chicago Section, American Chemical Society** American Chemical Society.

Chicago Section,1911 *Reduction of Hazardous Waste from High School Chemistry Laboratories* ,1989 Journal of Applied Microscopy and Laboratory Methods L. B. Elliott,1898 **Proceedings of the American Chemical Society**
American Chemical Society,1904 **Journal of the American Chemical Society** ,1904 **Review of American Chemical Research** ,1904

Melting Point Lab For High School Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Melting Point Lab For High School**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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