

Name Da'miya RobinsonDate 10-24-23

Draw place value disks to represent the value of the following expressions.

1. $5 \times 2 = 10$

5 times 2 ones is 10 ones.

thousands	hundreds	tens	ones

2. $5 \times 20 = 100$

5 times 2 tens is 10 tens

thousands	hundreds	tens	ones

3. $5 \times 200 =$ _____

5 times _____ is _____

thousands	hundreds	tens	ones

4. $5 \times 2,000 =$ _____

_____ times _____ is _____

thousands	hundreds	tens	ones

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655
 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
 one mole of HCl would be fully neutralized by one mole of NaOH

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ aq $\text{C}_6\text{H}_4\text{COONa}$ aq H_2O l the point of equivalence is the point of titration when the number of moles of naoh added is equal to the number of moles of khp nb in the

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