

CHEMISTRY A LEVEL MARK SCHEME

ALDEHYDES AND KETONES

1.	<p>- D – It reacts with KCN in a nucleophilic substitution reaction. (1)</p> <p>EXPLANATION:</p> <p>A is clearly wrong. Benzene rings undergo electrophilic substitution and this is a cyclohexadiene ring. It could undergo electrophilic addition with HBr, so really, the fault here is with the word “substitution.”</p> <p>B is also wrong. NaBH₄, also referred to as tetrahydridoborate (III), is used in reduction reactions – if you were trying to make this molecule into an alcohol, you’d be all set! Nucleophilic addition can also occur using aldehydes, but commonly uses HCN. Nucleophilic addition-elimination (what the question mentions) however, is a completely different ballpark – it occurs when nucleophiles react with acyl chlorides and acid anhydrides.</p> <p>C is also wrong. Ethanollic KOH is used for elimination, and this molecule does have a C-Br bond. However, there is no carbon-hydrogen bond adjacent to the C-Br bond – both of the adjacent carbons have used up all of their bonds!</p> <p>D is the right answer. Because of the C-Br bond, a negative cyanide ion can attack the C-Br bond and replace the bromine ion.</p>
2.	<p>- A – both can be formed by the alkaline hydrolysis of esters. (1)</p> <p>EXPLANATION:</p>

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Dilke,Thomas Kibble Hervey,William Hepworth Dixon,Norman Maccoll,Vernon Horace Rendall,John Middleton Murry,1896
Iron & Coal Trades Review ,1921 Vol 115 includes Diamond jubilee issue 1867 1927 *The Subject Index to Periodicals*
,1919 **APAIS 1994: Australian public affairs information service** , **Report of Her Majesty's Civil Service**
Commissioners Great Britain. Civil Service Commission,1897 *The Engineer* ,1894 *Modern English Biography*
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