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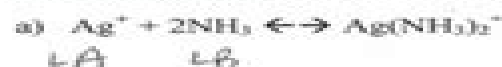
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### Review Sheet

1) In the following equations, label each species as an acid or a base. Show the conjugate acid-base pairs.



2) In the following reactions, identify the Lewis acid and the Lewis base.



3) For the following reaction, decide which species (reactants or products) are favored at the completion of the reaction. Explain.



4) Which member of each of the following pairs is the stronger acid? Explain.  *$\text{HSO}_4^-$  is stronger acid, therefore reactants are favored.*



Acid strength decreases  
down a group (except 17)



Acid strength increases  
across a period



There are more  
acidic hydrogens



The more oxygens an acid has,  
the more acidic the hydrogen are  
(inductive effect).



Acid strength decreases down  
a group (except 17).

# Key Idea 5 Standard 4 Review Sheet Answer Sheet

**Yijin Wang**



**Key Idea 5 Standard 4 Review Sheet Answer Sheet:**

## Decoding **Key Idea 5 Standard 4 Review Sheet Answer Sheet**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Key Idea 5 Standard 4 Review Sheet Answer Sheet**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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