## Chapter 7

## Combustion Processes in Engines

7.1) Compare the laminar flame speed s<sub>l</sub> of isooctane, propane, and methanol for P = 20 atm, T<sub>u</sub> = 600 K, f = 0.05, and φ = 0.9, 1.0, 1.1, and 1.2. At what equivalence ratios are the laminar flame speeds maximum for these fuels?

The laminar flame speed correlation (Equation 7.1) is

$$s_i = s_{i,0} \left(\frac{T_u}{T_{u,0}}\right)^a \left(\frac{P}{P0}\right)^b (1 - 2.1f)$$

The reference flame speed is

$$s_{0,0} = c_1 + c_2 (\phi - \phi_0)^2$$

Where  $c_1, c_2$ , and  $\phi_0$  are from Table 7.1

% This program computes laminar flame speed sl(cm/s)
phi = 1.2;
Tu=600;
P=20;
f=0.06;
a= 2.18-0.8\*(phi-1);
b=-.016+0.22\*(phi-1);
c1=36.92;
c2=-140.51;
phio=1.11;
slo=c1\*c2\*(phi-phio)\*2;
sl=slo\*(Tu/298)\*a\*P\*b\*(1-2.1\*f)

For  $T_v = 600$  K, f = 0.05, P = 20 atm, the following flame speed table is formed:

0	isooctane [cm/s]	propane	methanol
0.9	84.8	115.5	119.3
1.0	97.6	130.7	138.1
1.1	104.0	135.3	146.2
1.2	103.6	128.9	143.2

By differentiating Equation 7.1 with respect to  $\phi$  and setting it equal to zero, we find the laminar flame speed is maximized at slightly rich conditions at an equivalence ratio of  $\phi = \phi_0$ .

# **Manual On Combustion Engines**

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