



৪. কম-বেশি ১৫০ শব্দে যে-কোনো একটি প্রশ্নের উত্তর দাও :

৫

৪.১ “নদীকে এভাবে ভালোবাসিবার একটা কৈফিয়ত নদেরচাঁদ দিতে পারে।” — নদের-চাঁদ কী কৈফিয়ত দিতে পারে ? এভাবে ভালোবাসার কী পরিণতি হয় ?

২+৩

৪.২ “পরম্পরের প্রতি ভালোবাসার গল্প শুনে তাঁদেরও বুক ভরে গেল।” — কাদের ভালোবাসার কথা বলা হয়েছে ? পরম্পরের প্রতি তাদের ভালোবাসার পরিচয় দাও।

১+৪

৫. কম-বেশি ১৫০ শব্দে যে-কোনো একটি প্রশ্নের উত্তর দাও :



৫

৫.১ “নমি পুত্র পিতার চরণে,
করযোড়ে কহিলো —

পিতা ও পুত্রের পরিচয় দাও। কবিতা অবলম্বনে পিতা-পুত্রের কথোপকথন সংক্ষেপে লেখো।

১+৪

৫.২ “চিরচিহ্ন দিয়ে গেল তোমার অপমানিত ইতিহাসে।।” —

কাকে একথা বলা হয়েছে ? কীভাবে তার ‘অপমানিত ইতিহাসে’ চিরচিহ্ন দিয়ে গেল ?

১+৪

৬. কম-বেশি ১৫০ শব্দে যে-কোনো একটি প্রশ্নের উত্তর দাও :

৫

৬.১ “আমার মনে পড়ে প্রথম ফাউন্টেন কেনার কথা।” —

বক্তা কোথায় ফাউন্টেন কিনতে গিয়েছিলেন ?
তার ফাউন্টেন কেনার ঘটনাটি বিবৃত করো।



১+৪

৬.২ “এই দোষ থেকে মুক্ত না হলে বাংলা বৈজ্ঞানিক সাহিত্য সুপ্রতিষ্ঠিত হবে না।” —
কোন দোষের কথা বলা হয়েছে ? কীভাবে এই দোষ থেকে মুক্ত হওয়া সম্ভব বলে
লেখক মনে করেন ?

২+৩

৭. কম-বেশি ১২৫ শব্দে যে-কোনো একটি প্রশ্নের উত্তর দাও :

৪

৭.১ “তার সামান্য পরিচয় আজ দিয়ে রাখলাম।” —

কার উদ্দেশ্যে বক্তা এই কথা বলেছেন ?



তিনি কী ‘সামান্য পরিচয়’ দিয়ে রাখলেন ?

১+৩

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objectives to correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objects can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

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