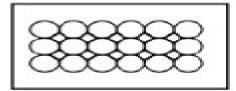
States of Matter and Phase Changes

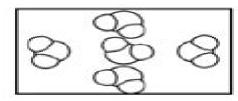
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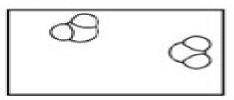
All forms of matter can exist in one of the four states of matter.

- Solids have a definite shape and volume. Atoms or molecules in a solid are held firmly in place, but can still vibrate. Solids have the least amount of energy.
- Liquids have a definite volume, but take the shape of their container. The atoms or molecules of a liquid are able to flow around each other.
- Gases contain molecules that fly at great speeds. Gases also take both the shape and volume of their container, assuming they are in one.
- Plasma is the highest-energy state of matter. In plasmas, the particles that make up the atoms and
 molecules escape and can move freely around.

The three diagrams below show the states of water. Label each as either a gas, liquid, or solid.





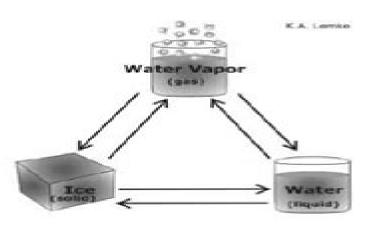


Phase Changes

Matter is most often converted from one state to another by adding or removing heat

- Vaporization converts from liquid to gas.
- Condensation converts from gas to liquid
- Freezing converts from liquid to solid.
- Melting converts from solid to liquid.
- Deposition converts from gas to solid.
- Sublimation converts from solid to gas.

Label each phase change on the diagram to the right



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Albert Cornelius Gaw

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