AP Chemistry Practice Test Chapter 8 and 9

Multiple Choice

Identify the choice that best completes the statement or answers the question.

13	Which pair of atoms should form the most polar bond	9
	a. F and B d. N	
		and N
	c. F and O	
23	Which pair of ions should form the ionic lattice with the highest energy?	
	a. Na ¹⁺ and Br ¹⁻ d. Li	1* and O2-
	b. Li ^{1*} and F ¹⁻ c. K	1+ and F1-
	c. Cs ¹⁺ and F ¹⁻	
 3.	The electrostatic force of attraction is greatest in which compound?	
	a. BaO d. M	
	b. MgO e. C.	aO
	c. CaS	
4.	4. Which molecule has the weakest bond?	
	a, CO d. N	
	b. O ₂ e. C	l ₂
	c. NO	
5.	5. For which species are octet resonance structures neces	
		eF ₂
	b. SO ₂ c. So	D ₄ -
	c. CO ₂	
 6.	6. How are the bonding pairs arranged in the best Lewis	
	a. O-O-O d. O	
	b. O=O-O e. O	=O=O
	c. O=O-O	
7.	7. Which species has the shortest bond length?	
	a. CN ¹ - d. St	O_1 :
	b. O ₂ e. O	0,
	e. SO ₂	
8.	8. Which species requires the least amount of energy to	remove an electron from the outermost energy level?
100	a. Na ^{1*} d. O	
		g ² *
	c. F1-	
9.	Which species has a valid non-octet Lewis structure?	
		eCl ₄
	b. SiF ₄ e. O	O ₂ -

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