

- Condensation polymerisation is the joining of two monomers with the elimination of small molecule such as water or HCl
  - Condensation polymerisation requires monomers with two different functional groups
  - A functional group on one monomer unit bonds to a different functional group on another monomer unit.
- Polyesters and polyamides are common condensation polymers.

### Polyesters

- An ester can be made from a carboxylic acid and an alcohol
- In forming the polyester, monomer units are bonded together by ester linkages and water is eliminated
- For polyester, the monomers have a carboxyl group and a hydroxyl group.
  - There are two common types of polyester:
    - Polyesters made by reacting two different types of monomer units:
      - One monomer is a dicarboxylic acid with two  $\text{-COOH}$  groups
      - The other monomer is a diol with two  $\text{-OH}$  groups
    - Polyesters made by reacting just one type of monomer unit containing  $\text{-COOH}$  and  $\text{-OH}$  groups

#### Polyesters from two types of monomer units

- During condensation polymerisation:
  - A bond forms between a hydroxyl group on the diol and a carboxyl group on a dicarboxylic acid
  - An ester linkage forms and water is eliminated.

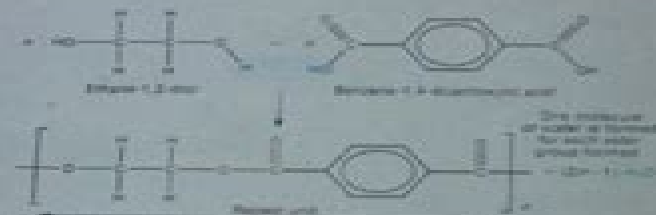


Figure 2 Formation of Terylene, a polyester

#### Polyesters from one type of monomer unit

- During condensation polymerisation:
  - A bond forms between a hydroxyl group on one molecule and a carboxyl group on a different molecules of the same type
  - As ester linkage forms and water is eliminated

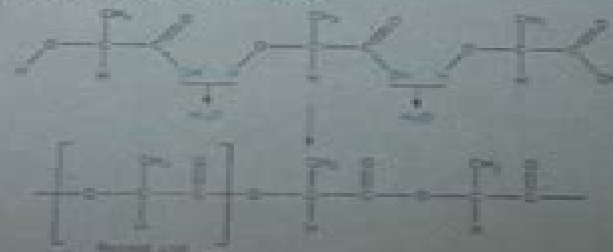


Figure 3 Condensation polymerisation of acid acid

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