

Solving Systems of Equations by Substitution

Date _____ Period _____

Solve each system by substitution.

$$\begin{aligned} 1) \quad y &= 7x - 10 \\ y &= -3 \end{aligned}$$

$$\begin{aligned} 2) \quad y &= -8 \\ y &= -2x - 12 \end{aligned}$$

$$\begin{aligned} 3) \quad y &= 6x \\ y &= 5x + 7 \end{aligned}$$

$$\begin{aligned} 4) \quad y &= 9x - 9 \\ y &= 9 \end{aligned}$$

$$\begin{aligned} 5) \quad y &= -4 \\ y &= x - 8 \end{aligned}$$

$$\begin{aligned} 6) \quad y &= 8x - 9 \\ y &= 7 \end{aligned}$$

$$\begin{aligned} 7) \quad y &= 6x - 14 \\ y &= -8x \end{aligned}$$

$$\begin{aligned} 8) \quad y &= 2x - 15 \\ y &= 5x \end{aligned}$$

Kuta System Of Equations Substitution Worksheet

Joacim Rocklöv



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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- \text{aq} + 2\text{Ag}^+ \text{aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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