

# Limiting Reactant Worksheet

- 1 Balance the following reaction:



- a Suppose 100 grams of sodium chloride reacts with 200 g of barium phosphate. What is the limiting reactant?

- b How many grams of excess reactant are left?

- 2 Acrylonitrile ( $\text{C}_3\text{H}_3\text{N}$ ) is the starting material for the production of a kind of synthetic fiber acrylic. It can be made from propylene ( $\text{C}_3\text{H}_6$ ) by reacting with nitric oxide (NO) as follows:



What mass of  $\text{C}_3\text{H}_3\text{N}$  can be made when 21.6 g of propylene react with 21.6 g of nitric oxide?

# Limiting Reagent And Percent Yield Worksheet Ib Chemistry

**Lifelique**



## **Limiting Reagent And Percent Yield Worksheet Ib Chemistry:**

*Limiting Reactant and Percent* Lifeliqe,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

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