LIMITING REACTANT LAB

MATERIALS: 250 mL beaker, 100 mL graduated cylinder, spoon or stirring rod, and a watch glass.

SAFETY PRECAUTIONS: Wear goggles during this experiment. Avoid skin contact with copper (II) chloride dihydrate. Wash your hands before leaving the laboratory.

PRELAB QUESTION:

Aluminum metal reacts with aqueous copper (II) chloride dihydrate (CuCl₂ · 2 H₂O) to form copper metal, aqueous aluminum chloride, and water. Write the balanced equation for this reaction, including state symbols.

PROCEDURE:

DAY 1

- Weigh out the copper (II) chloride dehydrate. Record the true mass of the hydrated compound used.
- Using the chart below, select your group number and the <u>approximate</u> masses of aluminum metal and copper (II) chloride dihydrate you'll be using in this activity:

Groups	Mass of Al (g)	Mass of CuCl2 · 2 H2O (g)
187813	1.00	2.00
288814	0.50	2.00
389	0.50	4.00
4 & 10	0.50	6.00
5 & 11	0.25	5.00
6 & 12	0.25	4.00

- Using a graduated cylinder, measure water (tap water is fine) and add it to your LABELED beaker containing the copper (II) chloride dehydrate. Stir until the salt dissolves completely. Keep adding water (maximum of 100 mL) until all salt is dissolved – record the amount of water used. Note the color of the solution.
- 4. Weigh out the appropriate mass of aluminum foil, record that mass, then crumple it up into a few small pieces (NOT TIGHTLY CRUMPLED) and add it to the copper (II) chloride dihydrate solution. Stir the solution gently and then allow it to sit undisturbed for 5-10 minutes. Write observations.

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Mocktime Publication

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Experimental Organic Chemistry Daniel R. Palleros, 2000-02-04 This cutting edge lab manual takes a multiscale approach presenting both micro semi micro and macroscale techniques The manual is easy to navigate with all relevant techniques found as they are needed Cutting edge subjects such as HPLC bioorganic chemistry multistep synthesis and more are presented in a clear and engaging fashion 15 PGT Chemistry Test Papers EMRS Mocktime Publication, EMRS Exam Teachers PGT Chemistry Test Papers 15 Practice Papers Tier 1 Eklavya Model Residential Schools as per Official Exam Practical Chemistry for CSEC Norman Lambert, 1987-03-30 Practical Chemistry is a unique practice Pattern and Syllabus book for CXC It provides a wealth of revision exercises and a guide to all the detailed experimental work covered in the CXC Chemistry syllabus Section A Practical guidance for teachers and classes perform **Research Based Undergraduate** Science Teaching Dennis W. Sunal, Cynthia S Sunal, Emmett L. Wright, Cheryl L. Mason, Dean Zollman, 2014-07-01 Research in Science Education RISE Volume 6 Research Based Undergraduate Science Teaching examines research theory and practice concerning issues of teaching science with undergraduates This RISE volume addresses higher education faculty and all who teach entry level science The focus is on helping undergraduates develop a basic science literacy leading to scientific expertise RISE Volume 6 focuses on research based reforms leading to best practices in teaching undergraduates in science and engineering The goal of this volume is to provide a research foundation for the professional development of faculty teaching undergraduate science Such science instruction should have short and longterm impacts on student outcomes The goal was carried out through a series of events over several years The website at http nseus org documents materials from these events The international call for manuscripts for this volume requested the inclusion of major priorities and critical research areas methodological concerns and results of implementation of faculty professional development programs and reform in teaching in undergraduate science classrooms In developing research manuscripts to be reviewed for RISE Volume 6 researchers were asked to consider the status and effectiveness of current and experimental practices for reforming undergraduate science courses involving all undergraduates including groups of students who are not always well represented in STEM education To influence practice it is important to understand how researchbased practice is made and how it is implemented The volume should be considered as a first step in thinking through what reform in undergraduate science teaching might look like and how we help faculty to implement such reform *Green Chemistry Education* Mark Anthony Benvenuto, Larry Kolopajlo, 2018-12-17 The greening of industry processes i e making them more sustainable is a popular and often lucrative trend which has emerged over recent years The 4th volume of Green Chemical Processing considers sustainable chemistry in the context of education and explores didactic approached The American Chemical Society s 12 Principles of Green Chemistry are woven throughout this text as well as the series to which this book belongs Forensics in Chemistry Sara McCubbins, Angela Codron, 2012 Forensics seems to have the unique ability to maintain

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