

INORGANIC CHEMISTRY

451. Nitrogen can be obtained from air by removing
 (a) Oxygen (b) Hydrogen (c) Carbon dioxide (d) Both (a) and (c)
452. Which of the following carbide is a methanide?
 (a) Be_2C (b) CaC_2 (c) Al_2C_3 (d) Mg_2C_3
453. Caro's acid is
 (a) H_2SO_3 (b) $\text{H}_2\text{S}_2\text{O}_5$ (c) H_2SO_4 (d) $\text{H}_2\text{S}_2\text{O}_8$
454. Inorganic benzene is
 (a) $\text{B}_3\text{H}_3\text{N}_3$ (b) $3\text{BH}_3\text{NH}_3$ (c) $-\text{B}_3\text{H}_6\text{N}_3-$ (d) $\text{H}_3\text{B}_3\text{N}_3$
455. For the reaction, $\text{P(White)} \longrightarrow \text{P(Red)}$; $\Delta H = -18.4 \text{ kJ}$. It follows that
 (a) Red P is readily formed from white P
 (b) White P is readily formed from red P
 (c) White P can not be converted to red P
 (d) White P can be converted into red P and red P is more stable
456. Nitrogen dioxide cannot be obtained by heating
 (a) KNO_3 (b) $\text{Pb(NO}_3)_2$ (c) $\text{Cu(NO}_3)_2$ (d) AgNO_3
457. Reactivity of borazole is greater than that of benzene because:
 (a) Borazole is a non-polar compound
 (b) Borazole is a polar compound
 (c) Borazole is an electron deficient compound
 (d) Of localized electrons in it
458. When orthophosphoric acid is heated to 600°C , the product formed is
 (a) Phosphine, PH_3 (b) Phosphorus pentoxide, P_2O_5
 (c) Phosphorus acid, H_3PO_3 (d) Metaphosphoric acid, HPO_3
459. $\text{P}_4 + 3\text{NaOH} + 3\text{H}_2\text{O} \rightarrow \text{A} + 3\text{NaH}_2\text{PO}_2$, here, 'A' is
 (a) NH_3 (b) PH_3 (c) H_2PO_4 (d) H_3PO_4
460. SO_2 in the solid phase exists as
 (a) linear chain or cyclic trimer (b) planar triangular unit
 (c) square planar unit (d) none of these
461. Sulphur does not exist as S_2 molecule because
 (a) It is less electronegative (b) Does not form pi-pi bond
 (c) It has ability to exhibiting catenation (d) Has tendency to show variable oxidation state
462. Which of the following has S-S BOND?
 (a) $\text{H}_2\text{S}_2\text{O}_8$ (b) $\text{H}_2\text{S}_2\text{O}_7$ (c) $\text{H}_2\text{S}_2\text{O}_6$ (d) Mustard gas
463. Which of the following is correct order of stability order?
 (a) $\text{Pb}^{2+} > \text{Pb}^{4+}$ (b) $\text{Pb}^{4+} > \text{Pb}^{2+}$ (c) $\text{Si}^{2+} > \text{Si}^{4+}$ (d) $\text{Sn}^{4+} > \text{Sn}^{2+}$
464. Quartz is a crystalline variety of
 (a) Si (b) SiO_2 (c) Na_2SiO_3 (d) SiC
465. The equivalent weight of phosphoric acid in the reaction
 $\text{NaOH} + \text{H}_3\text{PO}_4 \longrightarrow \text{NaH}_2\text{PO}_4 + \text{H}_2\text{O}$
 (a) 25 (b) 49 (c) 59 (d) 98

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NARAYAN CHANGDER



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1. Understanding the eBook Mcq Inorganic Chemistry Pearson
 - The Rise of Digital Reading Mcq Inorganic Chemistry Pearson
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcq Inorganic Chemistry Pearson
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcq Inorganic Chemistry Pearson
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcq Inorganic Chemistry Pearson
 - Personalized Recommendations
 - Mcq Inorganic Chemistry Pearson User Reviews and Ratings
 - Mcq Inorganic Chemistry Pearson and Bestseller Lists
5. Accessing Mcq Inorganic Chemistry Pearson Free and Paid eBooks
 - Mcq Inorganic Chemistry Pearson Public Domain eBooks
 - Mcq Inorganic Chemistry Pearson eBook Subscription Services
 - Mcq Inorganic Chemistry Pearson Budget-Friendly Options
6. Navigating Mcq Inorganic Chemistry Pearson eBook Formats
 - ePub, PDF, MOBI, and More
 - Mcq Inorganic Chemistry Pearson Compatibility with Devices
 - Mcq Inorganic Chemistry Pearson Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcq Inorganic Chemistry Pearson
 - Highlighting and Note-Taking Mcq Inorganic Chemistry Pearson
 - Interactive Elements Mcq Inorganic Chemistry Pearson
8. Staying Engaged with Mcq Inorganic Chemistry Pearson

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcq Inorganic Chemistry Pearson
9. Balancing eBooks and Physical Books Mcq Inorganic Chemistry Pearson
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcq Inorganic Chemistry Pearson
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mcq Inorganic Chemistry Pearson
- Setting Reading Goals Mcq Inorganic Chemistry Pearson
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mcq Inorganic Chemistry Pearson
- Fact-Checking eBook Content of Mcq Inorganic Chemistry Pearson
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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