

Ideal Gas Law

The ideal gas law is the equation of state of an ideal gas that relates pressure, volume, quantity of gas, and absolute temperature.

$$PV = nRT$$

P = pressure

V = volume

n = number of moles of gas

R = ideal gas constant

T = absolute temperature

Although it applies to ideal gases, it approximates the behavior of many real gases.

Mastering Physics Ideal Gas Law Answers

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University Physics: Australian edition Hugh D Young, Roger A Freedman, Ragbir Bhathal, 2010-08-04 This book is the product of more than half a century of leadership and innovation in physics education When the first edition of University Physics by Francis W Sears and Mark W Zemansky was published in 1949 it was revolutionary among calculus based physics textbooks in its emphasis on the fundamental principles of physics and how to apply them The success of University Physics with generations of several million students and educators around the world is a testament to the merits of this approach and to the many innovations it has introduced subsequently In preparing this First Australian SI edition our aim was to create a text that is the future of Physics Education in Australia We have further enhanced and developed University Physics to assimilate the best ideas from education research with enhanced problem solving instruction pioneering visual and conceptual pedagogy the first systematically enhanced problems and the most pedagogically proven and widely used online homework and tutorial system in the world Mastering Physics *The Software Encyclopedia*, 1988 *Government Reports Announcements & Index*, 1988 **The Electrician**, 1939 **The Electrical Journal**, 1938 *The Oil Weekly*, 1944

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