

scattering is incorrect?

- (i) charged object
- (ii) small in the impact parameter
- (iii) both interaction with the water molecule that

are deflected through 45° or more by a $0.1 \mu\text{m}$ gold

MILIR ISLAMIA
JMI

placed in the H_2 emission spectrum from hydrogen

of an atomic spectral line of wavelength 500 nm

correctly describes a system of two identical, non-
g' g'?

- (A) $\frac{1}{2}[\psi(1)\psi(2) + \psi(2)\psi(1)]$
- (B) $\frac{1}{2}[\psi(1)\psi(2) - \psi(2)\psi(1)]$
- (C) $\frac{1}{2}[\psi(1)\psi(2) + \psi(2)\psi(1)]$
- (D) $\frac{1}{2}[\psi(1)\psi(2) - \psi(2)\psi(1)]$

called by the name symbol $(\text{H})_{\text{H}}$. What are the
atomic structure?

65. Which of the following statements is

- (A) Molecular rotational states are all
- (B) The selection rule for molecular
- (C) Hydrogen molecule (H) is not in
- (D) Transitions between low lying

DRP JMI

66. A 10 kW broadcasting antenna emits
photons are emitted per second are

- (A) 1.5×10^{23}
- (B) 1.7×10^{23}
- (C) 1.8×10^{23}
- (D) 2×10^{23}

67. A photon is incident on an electron at
observed to be 3 times the wavelength of
the wavelength of the incident photon is

- (A) $1 \times 10^{-10} \text{ m}$
- (B) $2 \times 10^{-10} \text{ m}$
- (C) $4 \times 10^{-10} \text{ m}$
- (D) $5 \times 10^{-10} \text{ m}$

68. The de Broglie wavelength of a particle
of length L , is

- (A) L
- (B) $2L$
- (C) L
- (D) $L/2$

69. The operator $\left(\frac{\partial}{\partial x} + x\right)\left(\frac{\partial}{\partial x} + x\right)$ is eq.

- (A) $\frac{\partial^2}{\partial x^2} - x^2$
- (B) $\frac{\partial^2}{\partial x^2} + x^2 + 1$
- (C) $\frac{\partial^2}{\partial x^2} - 2x\frac{\partial}{\partial x} + 1$
- (D) $\frac{\partial^2}{\partial x^2} - 2x\frac{\partial}{\partial x} - x^2$

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D Kirk



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