
Physical Setting Chemistry Regents January 2025 Questions And Answers 2025 Actual Exam!!

At standard pressure, compare the strength of IMF in NO_2 (g) and N_2 (g) -
At STP, * NO_2 has stronger IMF than N_2 *.

Based on Table F, which equation represents a saturated solution having
the lowest concentration of Cl^- ions?

a) $\text{NaCl} \rightarrow \text{Na}^+ + \text{Cl}^-$

January 24 2013 Chemistry Regents Answers

**Kaplan, Kaplan Educational Center
Staff**



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