NATURAL SCIENCES

Gr. 9 - LESSON PLAN - MATTER & MATERIALS

	TOPIC: COMPO	DUNDS	
Sub-topic: Names of Compounds		Duration: 1 Hour	
	CONTENT & CONCEP	TS (CAPS p63)	
KEY CONCEPTS			
 Chemical 	reaction		
Chemical	formula		
Names of	compounds according to their element	S.	
Specific Aims:	Specific Aim 1: 'Doing Science'		X
	Specific Aim 2: 'Knowing the subject of	ontent and making connections'	X
	Specific Aim 3: 'Understanding the uses of Science'		

LESSON OBJECTIVES:

- 1. Learners will be able to write the names of compounds as indicated on page 63 of the CAPS Document.
- 2. Learners will be able to make models of different compounds

RESOURCES REQUIRED:

- Periodic Table Poster
- Siyavula Sasol Inzalo (Learner book) page 163

TEACHING & LEARNING ACTIVITY:

Introduction

Briefly review and revise concepts dealt with in Grade 8, focusing on compounds

As each element has a unique name for example:

- Hydrogen
- Lithium

Remember the first 20 elements learnt in grade 8. How did you memorize them? Do you remember Mnemonics...

Hill - H; He - He; Lies - Li; Because - Be; Boys - B; Can - C; Not - N; Operate - O; Fireplaces - F;

New - No: Nation - No: Might - Mg: Also - Al: Sign - Si: Peace - P: Security - S: Clause - Cl

A - Ar; King - K; Can - Ca.

LEARNERS MAY RECITE THIS MINEMONIC AS A WAY OF MEMORIZING THE FIRST 20 ELEMENTS.

The teacher provides the following explanation to the learners

Each compound has a unique name. How is a compound formed?

- A compound is formed when TWO or more different elements combine chemically.
- Each atom in a compound are held by chemical bonds
- Compounds are formed during chemical reaction (when elements/ componds combine to form new substances)
- Each compound can be written as a **formula**, for example. H-O: NaCl: CuSO₄

What is a formula?

- This is a notation that shows which elements the compound is made up of.
- It indicates the total number of atoms in one unit of the compound
- It indicates the ratio in which atoms of different elements are bonded to each other
- It gives the total number of atoms in one unit of a compound.

For example, the formula for Copper Sulphate is CuSO.

Natural Sciences Caps Lesson Plan Grade 9

National Education Association of the United States **Natural Sciences Caps Lesson Plan Grade 9:**

This book delves into Natural Sciences Caps Lesson Plan Grade 9. Natural Sciences Caps Lesson Plan Grade 9 is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Natural Sciences Caps Lesson Plan Grade 9, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Natural Sciences Caps Lesson Plan Grade 9
 - Chapter 2: Essential Elements of Natural Sciences Caps Lesson Plan Grade 9
 - o Chapter 3: Natural Sciences Caps Lesson Plan Grade 9 in Everyday Life
 - o Chapter 4: Natural Sciences Caps Lesson Plan Grade 9 in Specific Contexts
 - ∘ Chapter 5: Conclusion
- 2. In chapter 1, this book will provide an overview of Natural Sciences Caps Lesson Plan Grade 9. The first chapter will explore what Natural Sciences Caps Lesson Plan Grade 9 is, why Natural Sciences Caps Lesson Plan Grade 9 is vital, and how to effectively learn about Natural Sciences Caps Lesson Plan Grade 9.
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- 6. In chapter 5, this book will draw a conclusion about Natural Sciences Caps Lesson Plan Grade 9. The final chapter will summarize the key points that have been discussed throughout the book.

 This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Natural Sciences Caps Lesson Plan Grade 9.

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