

Gas Stoichiometry Exam
2016-2017

Name KLANN KEY
Period _____ Date _____

Multiple Choice

Refer to the set of graphs below when answering the next two questions:

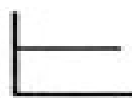
Graph #1



Graph #2



Graph #3



Graph #4



- B 1. Which of the graphs above is the best qualitative graph of pressure versus volume of nitrogen at constant temperature and number of moles?

A) Graph #1
B) Graph #2
C) Graph #3
D) Graph #4

→ INVERSE

- A 2. Which of the graphs above is the best qualitative graph of Kelvin temperature versus pressure of an ideal gas at constant volume and number of moles?

A) Graph #1
B) Graph #2
C) Graph #3
D) Graph #4

→ DIRECT

- D 3. Use the kinetic molecular theory of gases to predict what would happen to a closed sample of a gas whose temperature increased while its volume decreased.

→ MOVE FASTER, MORE COLLISIONS
A) The average kinetic energy of the molecules of the gas would decrease.
B) Its pressure would decrease.
C) The number of moles of gas would decrease.
D) Its pressure would increase.

- B 4. A hydrocarbon gas with the empirical formula CH_2 has a density of 1.3 g/L at 0°C and 1.00 atm. A possible formula for the hydrocarbon is:

A) CH_2
B) C_2H_4
C) C_3H_6
D) C_4H_8
E) C_5H_{10}

~~1.3 g/L~~

$$1.3 \text{ g/L} \times 22.4 \text{ L/mol} \approx 30 \text{ g/mol}$$

Modern Chemistry Gases Review Answer Key

Julia Schneider



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Table of Contents Modern Chemistry Gases Review Answer Key

1. Understanding the eBook Modern Chemistry Gases Review Answer Key
 - The Rise of Digital Reading Modern Chemistry Gases Review Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Gases Review Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Gases Review Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Gases Review Answer Key
 - Personalized Recommendations
 - Modern Chemistry Gases Review Answer Key User Reviews and Ratings
 - Modern Chemistry Gases Review Answer Key and Bestseller Lists

5. Accessing Modern Chemistry Gases Review Answer Key Free and Paid eBooks
 - Modern Chemistry Gases Review Answer Key Public Domain eBooks
 - Modern Chemistry Gases Review Answer Key eBook Subscription Services
 - Modern Chemistry Gases Review Answer Key Budget-Friendly Options
6. Navigating Modern Chemistry Gases Review Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Modern Chemistry Gases Review Answer Key Compatibility with Devices
 - Modern Chemistry Gases Review Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Chemistry Gases Review Answer Key
 - Highlighting and Note-Taking Modern Chemistry Gases Review Answer Key
 - Interactive Elements Modern Chemistry Gases Review Answer Key
8. Staying Engaged with Modern Chemistry Gases Review Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Gases Review Answer Key
9. Balancing eBooks and Physical Books Modern Chemistry Gases Review Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Gases Review Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Gases Review Answer Key
 - Setting Reading Goals Modern Chemistry Gases Review Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Gases Review Answer Key
 - Fact-Checking eBook Content of Modern Chemistry Gases Review Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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