# SCIENCE WORKSHEETS

NAME

Counting Atoms
Count the number of atoms and fill in the table.

H	I <sub>2</sub> 0	
H	2	
0	1	
TOTAL	3	

H <sub>2</sub> O <sub>2</sub>	
Н	2
0	2
TOTAL	4

CH <sub>4</sub>	
С	1
Н	4
TOTAL	5

3	H₂O	
Н	6	
0	3	
TOTAL	9	

3N	laCl
Na	3
Cl	3
TOTAL	6

6CO <sub>2</sub>		
C	6	
0	12	
TOTAL	18	

H <sub>2</sub> SO <sub>4</sub>		
Н	2	
S	1	
0	4	
TOTAL	7	

C6H12O6		
С	6	
Н	12	
0	6	
TOTAL	24	

2AgNO <sub>3</sub>		
Ag	2	
N	2	ľ
0	6	Ī
TOTAL	10	

# **Modern Chemistry Section 4 Counting Atoms**

**Rex M. Heyworth** 

#### **Modern Chemistry Section 4 Counting Atoms:**

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Modern Chemistry Nicholas D. Tzimopoulos, 1993 Introduction to Modern Inorganic Chemistry, 6th edition R.A. Mackay, 2017-12-21 This popular and comprehensive textbook provides all the basic information on inorganic chemistry that undergraduates need to know For this sixth edition the contents have undergone a complete revision to reflect progress in areas of research new and modified techniques and their applications and use of software packages Introduction to Modern Inorganic Chemistry begins by explaining the electronic structure and properties of atoms then describes the principles of bonding in diatomic and polyatomic covalent molecules the solid state and solution chemistry Further on in the book the general properties of the periodic table are studied along with specific elements and groups such as hydrogen the s elements the lanthanides the actinides the transition metals and the p block Simple and advanced examples are mixed throughout to increase the depth of students understanding This edition has a completely new layout including revised artwork case study boxes technical notes and examples All of the problems have been revised and extended and include notes to assist with approaches and solutions It is an excellent tool to help students see how inorganic chemistry applies to medicine the environment and biological topics

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